## Introduction to phytochemical pathways



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إعداد الطالبة: آلاء العتوم

Plant Metabolism

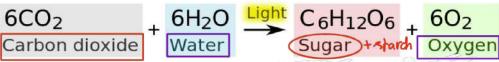
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 The biological reactions are essential to maintain life in living organisms and are known as primary metabolism.

 Plant convert sunlight energy to chemical energy, such as ATP, NADPH, by the mediation of chlorophyll in chloroplasts and synthesize sugars and starch from CO2 by using ATP and NADPH+.

 These carbohydrates are stored and used for differentiation and formation of plant tissues.

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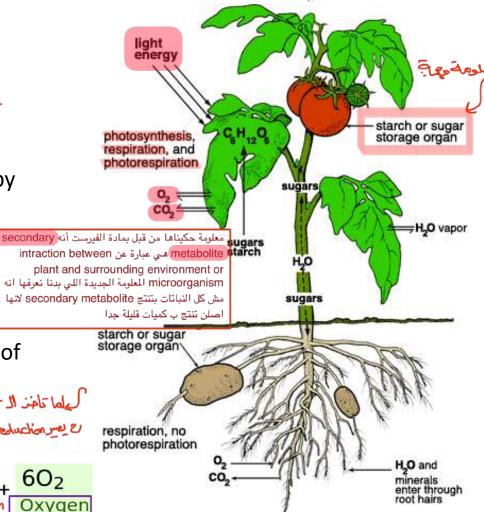
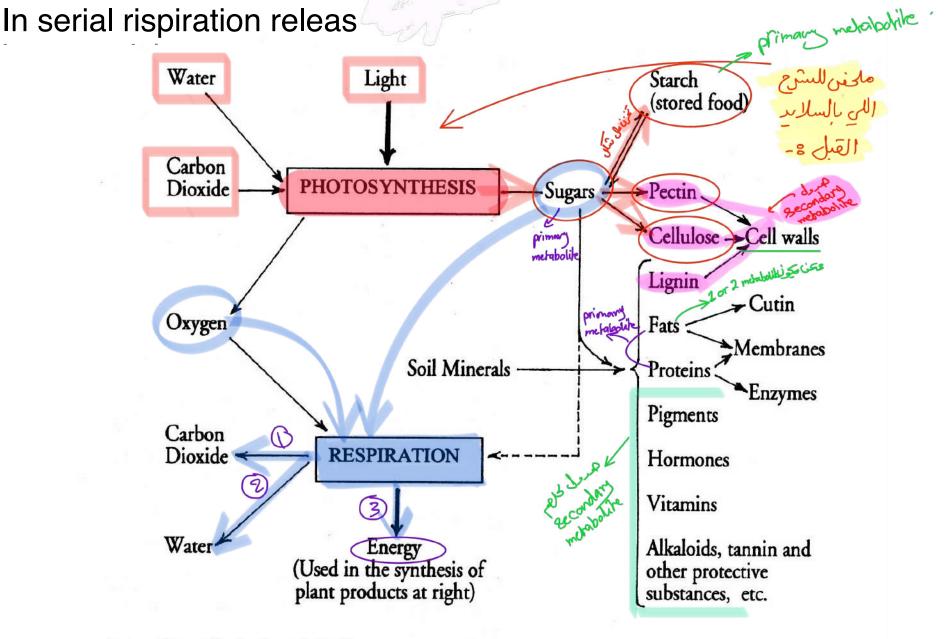


Figure 24. Photosynthesis, respiration, leaf water exchange, and translocation of sugar (photosynthate) in a plant.



An outline of plant metabolism

- A\* All organisms need to transform and interconvert a vast number of organic compounds to enable them to

  → live → grow → and reproduce.
- **B\*** They need to provide themselves with energy in the form of ATP.
- C\* They need a supply of building blocks to construct their own tissues.
- **D-**An integrated network of **enzymes** mediated and carefully regulated chemical reactions is used for this purpose collectively referred to as→**intermediary metabolism** and the pathways involved are termed→ **metabolic pathways.**
- **E\*** Some of the important molecules of life are:
- 1-Carbohydrates, 2- Proteins, 3-Fats and 4-Nucleic acids.

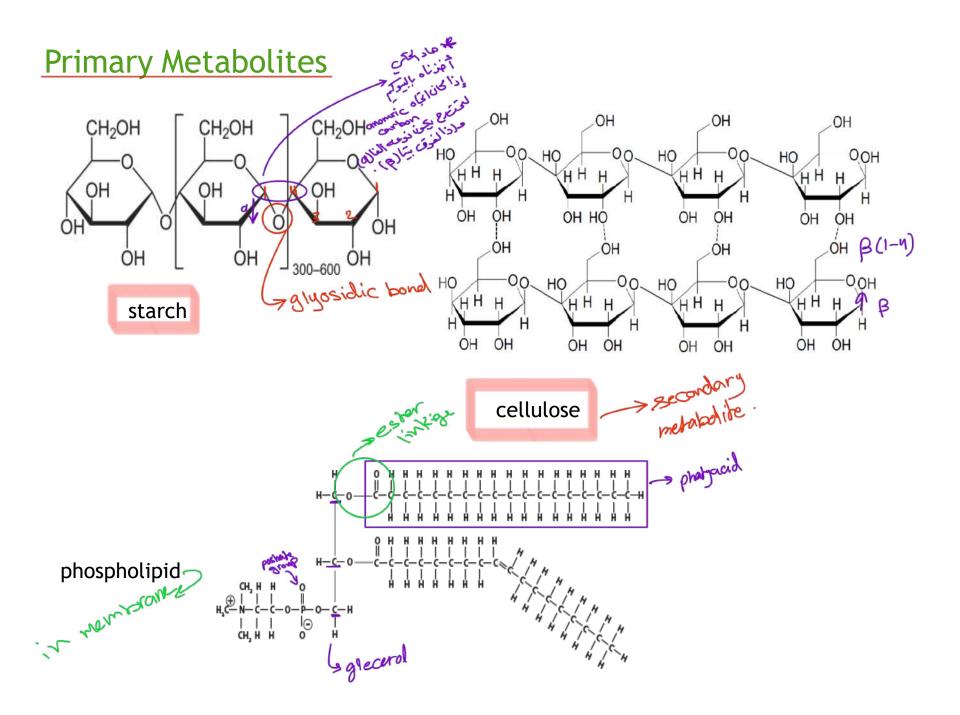
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## **Primary Metabolites**

- Primary metabolites are compounds produced by a living organism that are essential for its life and growth.
- The main primary metabolites are carbohydrates, proteins, nucleic acids, and lipids.

" fat" 11 le lo

- They are occurring in all plants (in all green plants).
- # Apart from fats, these are polymeric materials.
- \* Carbohydrates(CHO): are composed of <u>sugar units</u>.
- \* Proteins: are made up from amino acids
- \* Nucleic acids: are based on nucleotides.



## Plants and their metabolites

- Primary metabolites: Compounds produced during the fundamental pathways; essential for the plant life (H₂O, CO₂ → Photosynthesis): carbohydrates, fats, proteins, nucleic acids →primary activity of green plants
- Secondary metabolites: Not involved in the essential metabolism of the cell, but exert physiologic activity for the plant, its environment and human (phyto-hormons, plant-protection, plant-insect-, plant-animal interactions, source of drugs); produced through specific pathways with a limited distribution in nature; condition and species sensitive; most pharmacologically active natural products are 2<sup>ry</sup> metabolites

#### Relationship between primary and secondary metabolism:

- The processes and products of primary metabolism are similar in most organisms, while those of secondary metabolism are more specific.
- In plants, primary metabolism is made up of photosynthesis, respiration, etc., using CO<sub>2</sub>, H<sub>2</sub>O, and NH<sub>3</sub> as starting materials, and forming products such as glucose, amino acids, nucleic acids. These are similar among different species.
- Secondary metabolites are low molecular weight compounds that are <u>not</u> required for the growth of an organism, but are produced for adaptation for its specific functions in nature and are considered a phenotype of the organism.

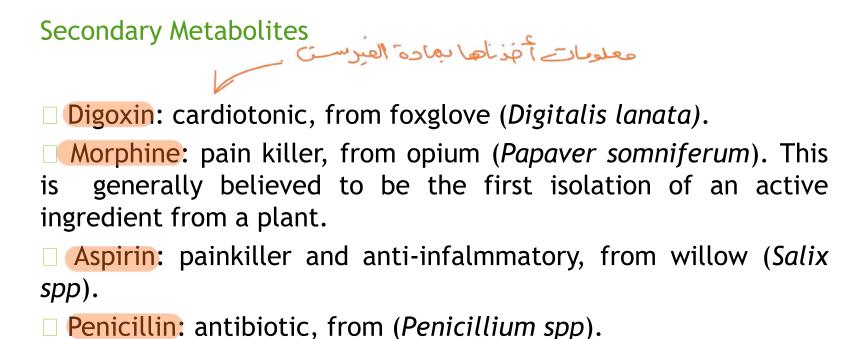
They are derived from primary metabolites via secondary metabolic

- Pathways
- They are more important than primary metabolites as a therapeutic agents.

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## **Secondary Metabolites**

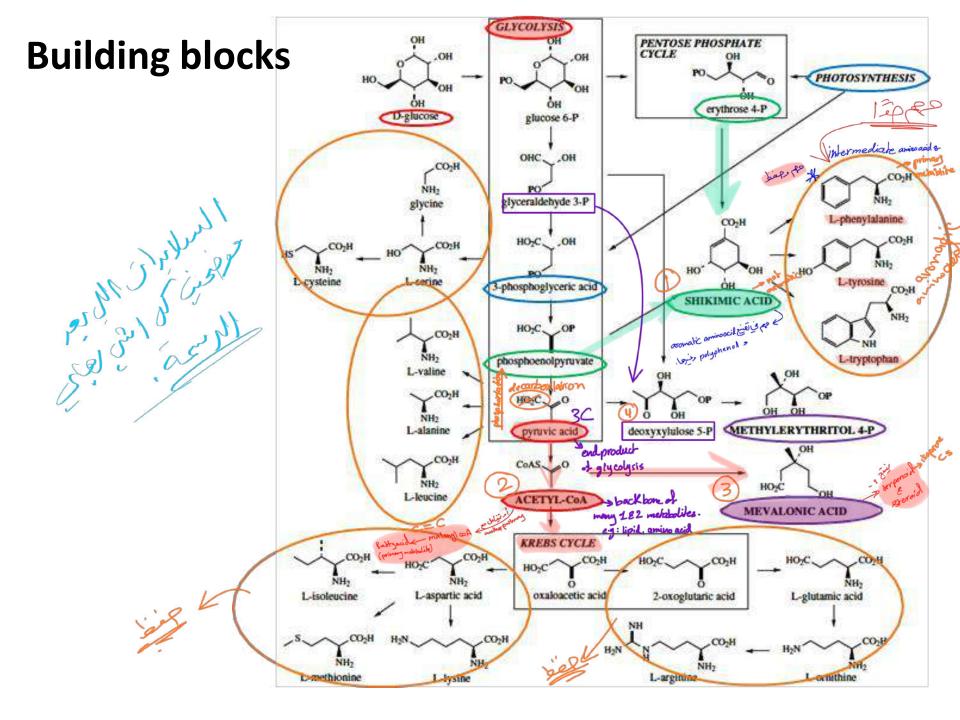
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## Functions of Secondary Compounds

- The most common roles for secondary compounds in plants are ecological roles that govern interactions between plants and other organisms.
- Many secondary compounds are brightly colored pigments like anthocyanin that color flowers red and blue. These attract pollinators and fruit and seed dispersers.
- Nicotine and other toxic compounds may protect the plant from herbivores and microbes.

 Other secondary compounds like <u>rubber</u> and <u>tetrahydrocannabinol</u> (THC) <u>from cannabis plants</u> have no known function in plants.



## **GLYCOLYSIS**

 Glycolysis: The final product is the pyruvuic acid  $(3 \text{ carbon}) \rightarrow \text{Then the pyruvate has } 2 \text{ options}$ :

• A -> Further degradation to acetyl-CoA (2 carbon) → it is a backbone of many primary and secondary metabolites

• B→ Shicimic acid pathway:

a glycolygic Pyruvic acid(phosphoenolpyruvate) combines with erythrose-4-P→ to form shikimic acid. from Krib-cycle

• \* So to form shikimic acid we need:

**A**-Calvin cycle

**B**-Glycolysis

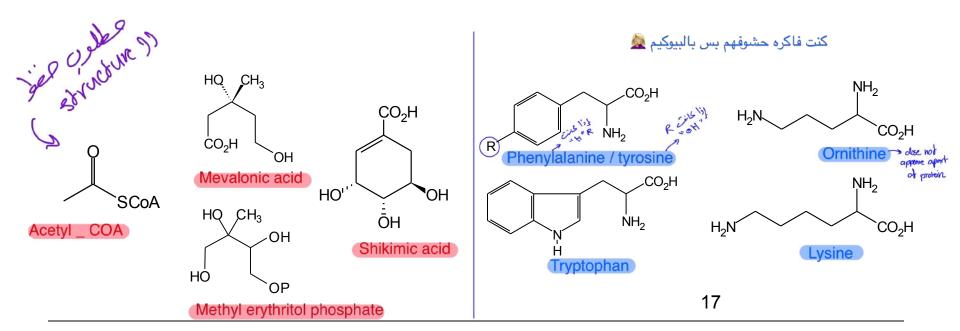
- Degradation of carbohydrates and sugars generally proceeds via the well characterized pathways, known as glycolysis and the kerbs / citric acid / tricarboxylic acid cycle, which release energy from the organic compounds by oxidative reactions.
- Oxidation of fatty acids from fats by the sequence called β-oxidation also provides energy

- Acetyl-CoA can go to one of the following pathways:
  - **1-**Tricarboxylic acid cycle → non aromatic plant acids(citric, succinic and malic acids)
- •2- To form primary metabolites as fatty acids depending on the enzymatic specificity: → Acetyl- CoA + malonyl-CoA will start the formation of fatty acids
- 3- Can enter the secondary metabolites formation depending on enzyme availability by **two pathways**:
- A- Mevalonic acid pathway -> terpenoids and steroids
- B- Acetate-malonate pathway > aromatic, monocyclic and polycyclic compounds(Anthraquinones, Prostaglandins, Fats, Marolide antibiotic(erythromycine), Tetracyclines
- \* Krebs cycle= tricarboxylic acid cycle=citric acid cycle

# AcetylCoA is formed by

- کل بهاد السلايد سيرح ۱۱ عهمالم الحكى ،
- 1→Oxidative decarboxylation of the glycolytic pathway product pyruvic acid.
- 2  $\rightarrow$ It is <u>also</u> <u>produc</u>ed by  $\rightarrow$ the  $\beta$ -oxidation of fatty acids, effectively reversing the process by which fatty acids are themselves synthesized from acetyl-CoA .
- ◆ Mevalonic acid → mevalonate(Mevalonat- acetate) →
- 1-Isoprene(Terpenoides)
- 2- Isoprene → Squalene → steroids
- Mevalonic acid is itself formed → from three molecules of acetyl-CoA, but the mevalonate pathway channels acetate into a different series of compounds
- than does the <u>acetate</u> pathway.
- Deoxyxylulose phosphate pathway:
- → arises from a <u>combination</u> of <u>two</u> glycolytic pathway <u>intermediates</u>, namely → <u>pyruvic acid</u> <u>and</u> → <u>glyceraldehyde 3-phosphate</u>.
- ◆Shikimic acid → shikimate pathway → 1- Aromatic amino acids , 2- phenols, 3- Alkaloids, 4- lignans, 5- peptides and 6-antibiotics(penicillins)
- **Shikimic acid** is produced from a combination of → **phosphoenolpyruvate**, a glycolytic pathwayintermediate, **and** → **erythrose 4-phosphate** from the pentose phosphate.
- pathway (see figure of the building blocks).

<u>Secondary metabolism</u>: refers to the biosynthesis, utilization and breakdown of smaller organic compounds found in the cell. These compounds, called **secondary metabolites**, arise from a set of intermediate building blocks: acetyl coenzyme A (acetyl-CoA), mevalonic acid (MVA) and methyl erythritol phosphate (MEP), shikimic acid, and the amino acids phenylalanine/tyrosine, tryptophan, ornithine and lysine.



## Constructing of natural products

- \* In addition to acetyl CoA, shikimic acid, mevalonic acid and deoxyxylulose phosphate, **other** building blocks based on **amino acids** are frequently employed in **natura**l products.
- 1-Peptides, proteins, alkaloids, and many antibiotics are derived→ from amino acids, and the origins of the most important amino acid components of these are briefly indicated in.
- Figure
- 2 -Intermediates from the <u>glycolytic</u> pathway and the <u>Krebs cycle</u> are used in constructing <u>many</u> of them.
- 3-Shikimate pathway → <u>aromatic amino a</u>cids
   → phenylalanine, tyrosine, and tryptophan.
- 4- Krebs cycle intermediates → Ornithine,
   a non-proteinamino acid, along with its homologue lysine,
  - → are important <u>alkaloid</u> precursors.

The building blocks

The hullding blocks 
$$H_{1}C^{-S} \underbrace{\hspace{0.5cm} CO_{2}H}_{NH_{2}} \Longrightarrow -x - CH_{1} \quad (x = 0, N, C)$$

$$\underbrace{\hspace{0.5cm} SCoA}_{NH_{2}} \underbrace{\hspace{0.5cm} CO_{2}H}_{NH_{2}} \Longrightarrow -x - CH_{1} \quad (x = 0, N, C)$$

$$\underbrace{\hspace{0.5cm} SCoA}_{SCoA} \underbrace{\hspace{0.5cm} CO_{2}H}_{SCoA} \Longrightarrow C - C G$$

$$\underbrace{\hspace{0.5cm} SCoA}_{SCoA} \Longrightarrow HO_{2}C \longrightarrow OH$$

$$\underbrace{\hspace{0.5cm} OH}_{NH_{2}} \longrightarrow OH \longrightarrow OH$$

$$\underbrace{\hspace{0.5cm} OH}_{NH_{2}} \longrightarrow OH$$

$$\underbrace{\hspace{0.5cm} OH}_{NH_{2}} \longrightarrow OH \longrightarrow OH$$

$$\underbrace{\hspace{0.5cm} OH}_{NH_{2}} \longrightarrow OH$$

$$\underbrace{\hspace{0.5cm} OH}_{NH_{2$$

#### C1:

- The C1 methyl building unit is supplied from
   Lmethionine and is introduced by
   → a nucleophilic substitution reaction.
- →In nature, the leaving group is enhanced by converting L-methionine into
   S-adenosylmethionine (SAM)

#### $\rightarrow$ C1: comment $\rightarrow$ Formation of SAM

- 1- This gives a <u>positively</u> charged <u>sulp</u>hur and facilitates
- → the <u>nucleop</u>hilic substitution (SN2) type mechanism
- 2-Thus, *O*-methyl and *N*-methyl linkages <u>may</u> be obtained
- → using <u>hydroxyl</u> and <u>amino functions</u> as <u>nucleop</u>hiles.
- 3- The methylenedioxy group (OCH2O)

## C1:

#### Alkylation reactions: nucleophilic substitution

# H<sub>2</sub>N H<sub>2</sub>O OH

 $CO_2H$ 

#### (b) O- and N-alkylation using SAM

S-adenosylmethionine

(SAM)

#### (c) C-alkylation using SAM

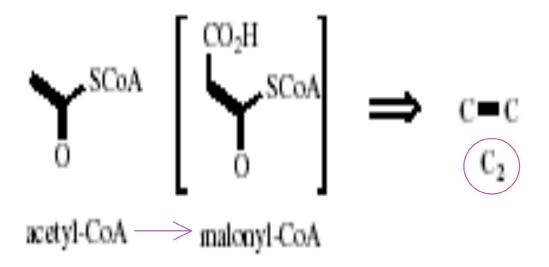
L-Met

carbonyl groups increase acidity and allow formation of enolate anion

neutral molecule is

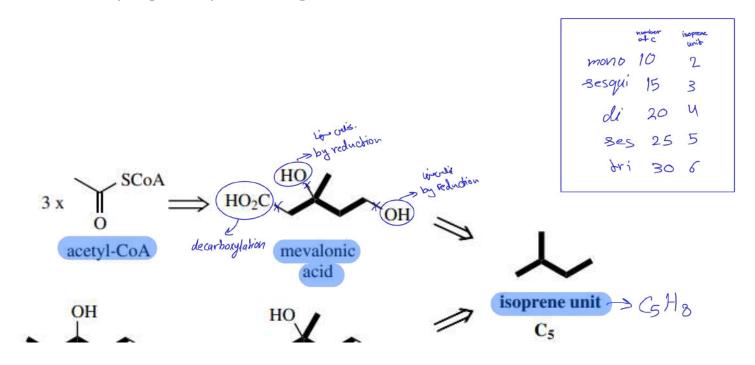
## C2:

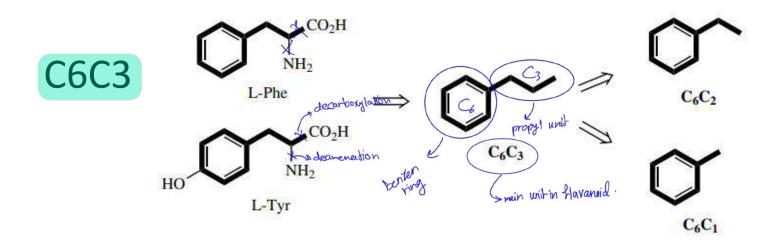
- 1- A two-carbon unit may be supplied by acetyl-CoA.
- 2- It <u>forms</u> part of a <u>long alkyl chain</u> (as in a <u>fatty</u> acid) or <u>may</u> be <u>part</u> of an <u>aromatic system</u> (e.g.phenols).
- 3- Of particular relevance is that in the latter examples, acetyl-CoA is first converted into the more reactive malonyl-CoA before its incorporation.



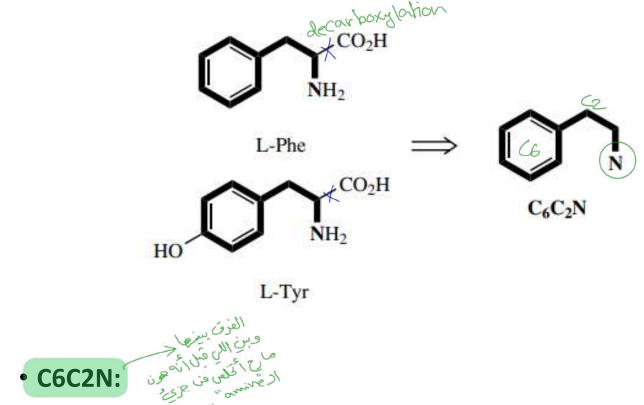
C5:

- **1-**The branched-chain C5 '<u>isoprene</u>' unit is a <u>feature</u> of compounds <u>formed</u>→from mevalonate.
- -Mevalona te itself is the product from three acetyl-CoA molecules, but only five of mevalonate's six carbons are used, the carboxyl group being lost.

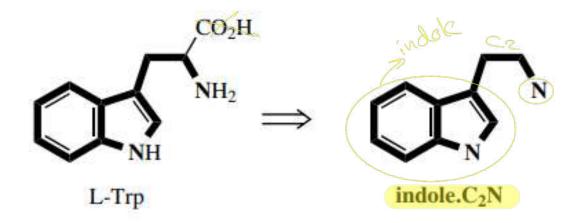




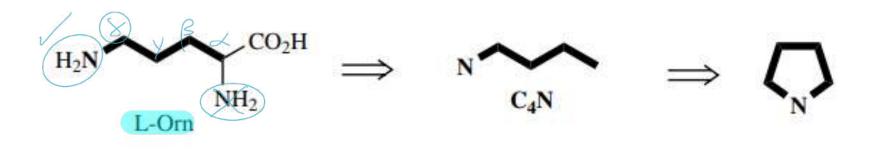
- 1-This refers to a phenylpropyl unit and is obtained
- →from the carbon skeleton of <u>either</u> L-<u>phenylalanine</u> or L-<u>tyrosine</u>, →two of the <u>shikimate</u>-derived <u>aromatic amino</u> acids.
- 2 -This, of course, requires loss of the amino group.
- 3 -The C3 <u>side-chain</u> may be <u>saturated</u> or unsaturated, and <u>may</u> be <u>oxygenated</u>.
- 4 Sometimes the side chain is cleaved, removing one or two carbons.
- → Thus, C6C2 and C6C1 units represent
- →modified shortened <u>forms</u> of the <u>C6C3 system</u>.



- 1-This building block is formed
  - → from <u>either</u> L-phenylalanine or L-tyrosine,
    - →Ltyrosine being by far the more common.
- 2-In the <u>elaboration</u> of this unit, the <u>carboxyl carbon</u> of the <u>amino</u> acid is <u>remove</u>d.

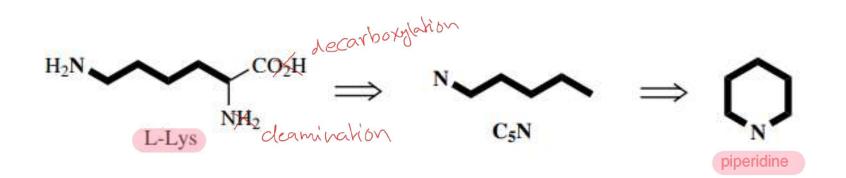


- (Indole.C2N):
- 1-It is formed → from the <u>aromatic amino acid</u>
  L-tryptophan(Shik. acid p.w )
- 2-This <u>indole</u>-containing system <u>can</u> undergo <u>decarboxylation</u> in a <u>similar</u> way to <u>L- phenylalani</u>ne and <u>L-tyrosi</u>ne
  - so <u>providing</u> → the remainder of the skeleton →as an <u>indole.C2N unit</u>.



#### • C4N:

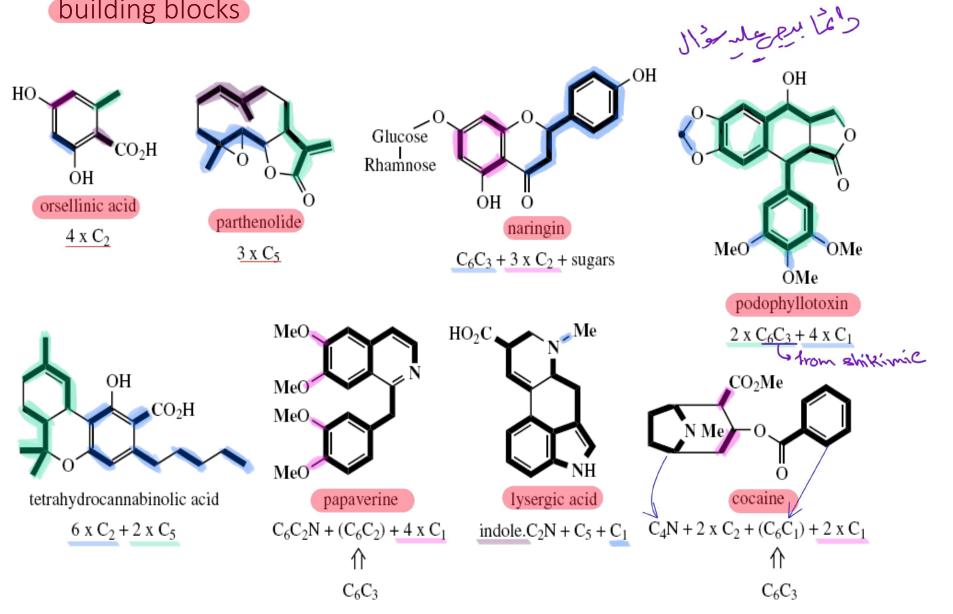
- 1-The C4N unit is usually found
   →as a heteroceylic <u>pyrrolidine</u> system.
- 2-It is produced from
  - → the non-protein amino acid L-ornithine.
  - **3**-In marked <u>contras</u>t to the <u>C6C2N</u> and <u>indole.C2N</u> units described above:
  - $\rightarrow$  <u>ornithine</u> <u>supplies</u> <u>not</u> its  $\alpha$ -amino nitrogen, <u>but</u> the <u>-amino</u> <u>nitrogen</u>.
- **4** -The <u>carboxylic</u> acid function and the  $\alpha$ -amino nitrogen
- →are <u>both</u> lost.



- C5N:
- 1-This is produced in **exac**tly the **same** way
- →as the C4N unit
- (piperidine)
- →<u>but</u> using <u>L-lysine</u> as <u>precu</u>rsor.

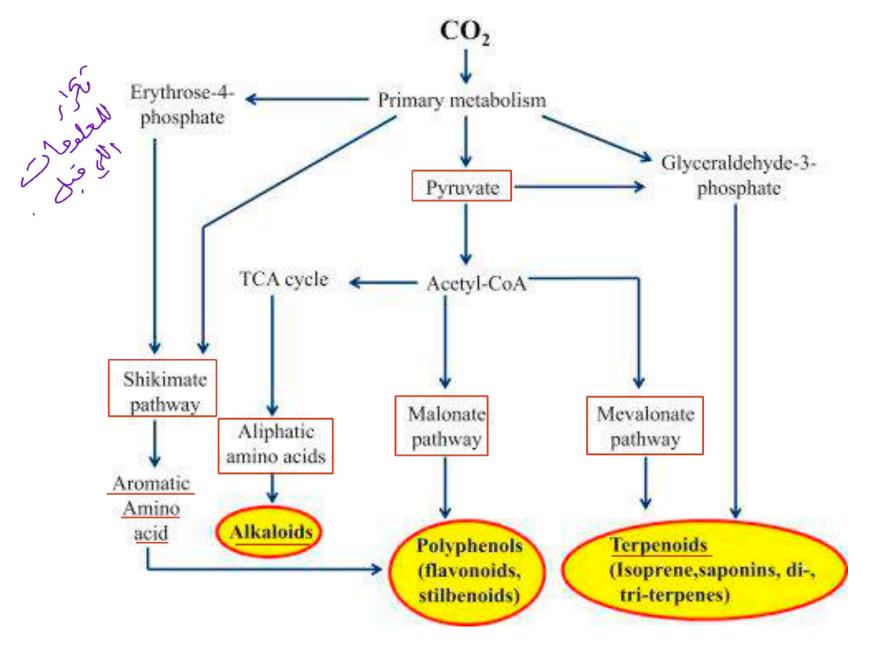
Examples of how compounds can be visualized as a combination of

building blocks



The different secondary metabolites are biosynthesized using 3 different pathways.

- 1) Acetate Malonate Pathway 2) Acetate –
   Mevalonate Pathway (terpenoids and steroids)
- 3) Shikimic acid pathway (aromatic amino acids and phenylpropanoids)
- Each of them is responsible for the production of certain classes of secondary metabolites.
  - 4- amino acid pathway.



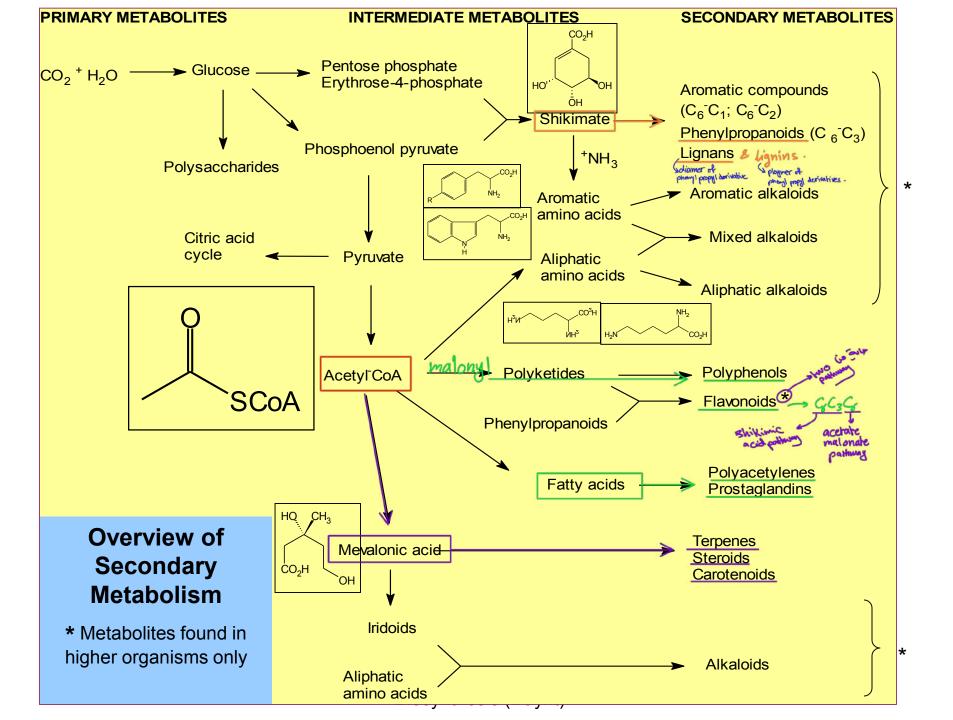
<u>Important secondary metabolites</u> formed <u>from</u> the <u>acetate pathway</u> include <u>phenols</u>, <u>prostaglandins</u>, and <u>macrolide antibiotics</u>, together with various <u>fatty acids</u> and derivatives at the primary/secondary metabolism interface.

Shikimic acid is produced from a combination of <u>phosphoenolpyruvate</u>, a glycolytic pathway intermediate, and <u>erythrose 4-phosphate</u> from the pentose phosphate pathway.

The shikimate pathway leads to a variety of phenols, cinnamic acid derivatives, lignans, and alkaloids.

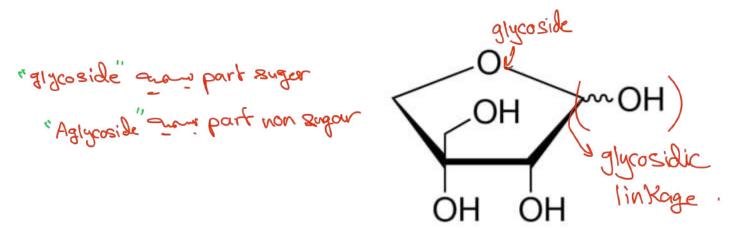
Mevalonic acid is itself formed from three molecules of acetyl-CoA, but the mevalonate pathway channels acetate into a different series of compounds than does the acetate pathway.

<u>Deoxyxylulose phosphate</u> arises from a combination of two glycolytic pathway intermediates, namely pyruvic acid and glyceraldehyde 3-phosphate.



## Plant Secondary Metabolites

- Secondary compounds are grouped into classes based on similar structures, biosynthetic pathways, or the kinds of plants that make them. The largest such classes are the alkaloids, terpenoids, and phenolics.
- Secondary compounds often occur in combination with one or more sugars. These combination molecules are known as glycosides. Usually the sugar is a glucose, galactose or rhamnose. But some plants have unique sugars. Apiose sugar is unique to parsley and its close relatives.



## Acetate malonate pathway

A-Non-aroamtic derivatives (polyacetylenes, منبونية
 polyynes)

B-Aromatic polyacetates (polyketides)

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### **Acetate-derived Natural Products**

Acetate formed from carbohydrate via pyruvic acid

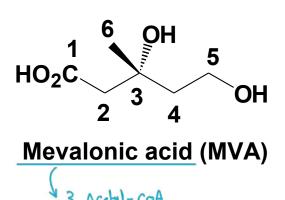
Produce large no. of imp. natural products (flavonoids, anthraquinones, macrolides, terpenes & steroids)

## 2 main routes originate with acetate pathway:

- 1. Acylpolymalonate pathway leading to FA & polyketides acelyl unite
- 2. Mevalonic acid pathway producing terpenes& steroids

## Acetate mevalonate pathway

- Isoprenoid compounds
- Terpenes (mono-, sesqui-, di-, tri-, tetra-, polyterpenes)
- Cardiac glycosides
- Saponin glycosides
- Steroids
- Vitamin D

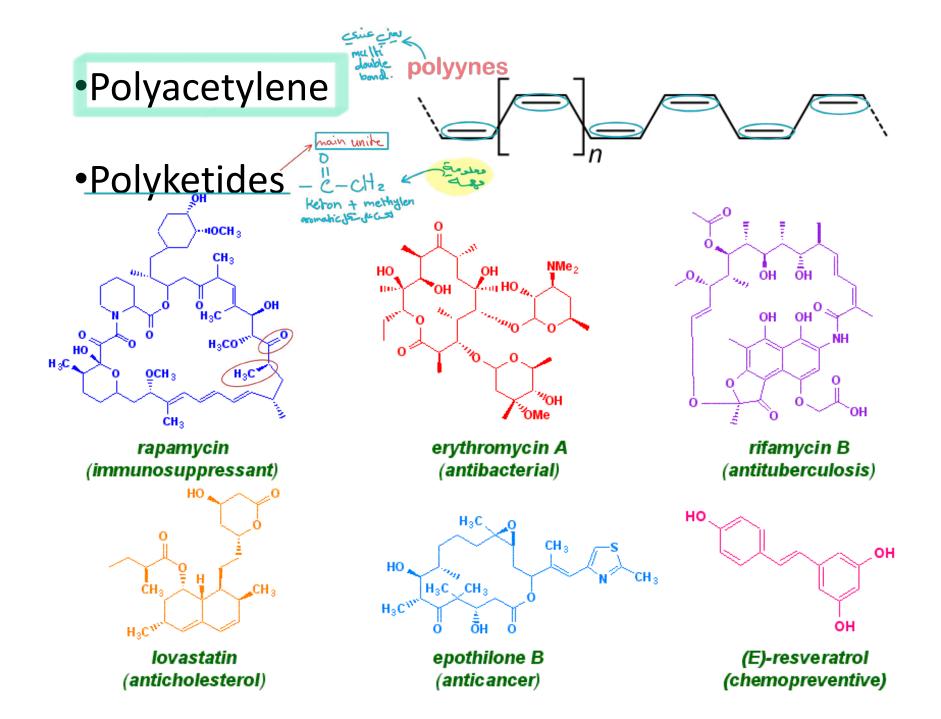


## Shikimic acid pathway

- Shikimic acid is a very important building block of phenolic substances, because it contains a ring and it's also important for the formation of three important amino acids:

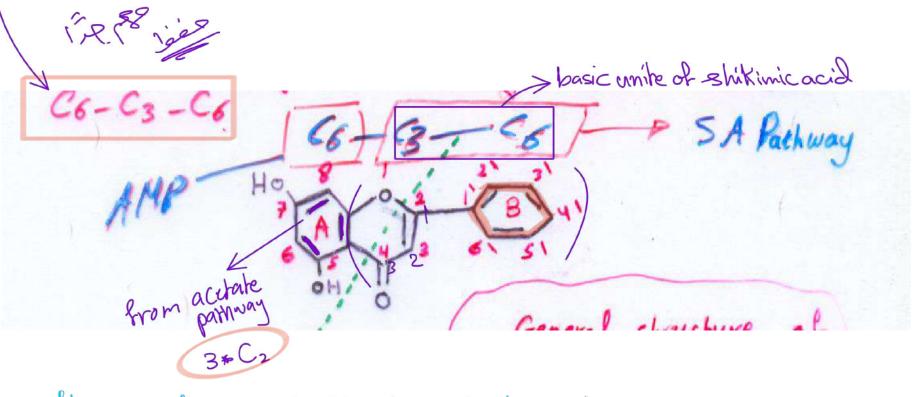
  Phenylalanine, Tyrosine and Tryptophan.
- Aromatic biosynthesis:
- Phenols
- Phenolic glycosides
- Phenyl-propane-derivatives
- Lignins
- Lignans
- Aromatic amino acids

**Shikimic Acid** 



## Flavanoids from <u>acetate malonate</u> (C6)+ <u>shikimic</u> acid(C6C3)

\*\*hint:in the record, the above sentences appear as "Flavanoids from mevalonic acid(C6)+ shikimic acid(C6C3)" and this is wrong sentence. The above is correct

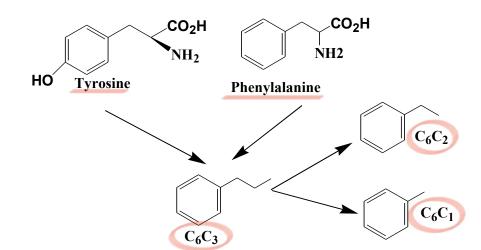


Havaroids \_ 1 ben Jen ving 2 other bentom ring

## Most important building blocks







## "Amino acid" pathway

Includes all biogenetic drugs containing "N".:

- رعية شرعها بالتفيل بالتشابر الهي بيده :
- -Cyanogenic glycosides
- -Glucosinolates
- -Alliines
- -Antibiotics from amino acid metabolism
- -Anticancer drugs from amino acid metabolism