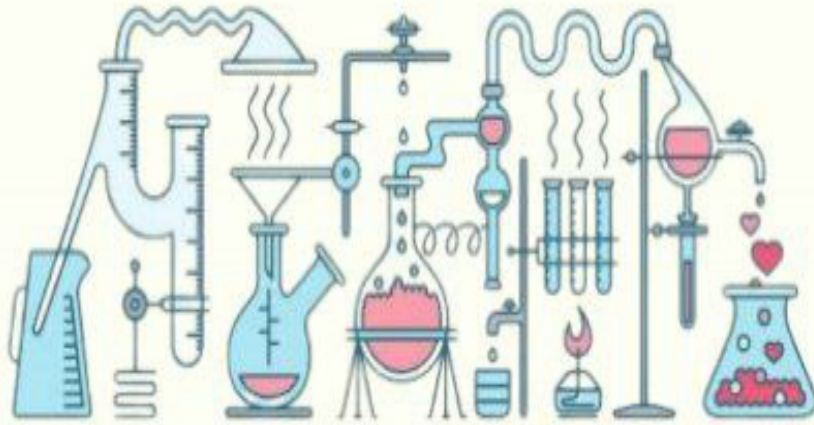


تفريغ تحليل آلي



اسم الموضوع: *Introduction*
Part 2

إعداد الصيدلاني /ة: *Sara Jaber*

Figures of Merit

Method Approved ☒ *علايات رياضية إذا توافقت الأرقام مع الشرط*

Performance Characteristics: Figures of Merit

How to choose an analytical method? How good is measurement?

or Judge

How well a calibration curve follows a straight line? - **Linearity**.

طابع من الجسيمات مع تغير التركيز
ببطل خطي

How reproducible? – **Precision**

Standards
actual / theory
من التجربة ← → من الشركة

How close to true value? - **Accuracy/Bias**

Mean علاقة بين العينات والـ

الانحراف

How small a difference can be measured? – **Sensitivity**

مستوى التحسس للـ methode

What range of amounts? - **Dynamic Range**

(أقل ← أعلى) يمكن قياسه بدقة

How much interference? - **Selectivity**

Selective
or not

FIGURES OF MERIT: Linearity

زيادة التركيز مع زيادة الجهد بشكل متزايد خطي

x → Conc. → ppm
y → Response of Instrument
استجابة الجهاز

- How well a calibration curve follows a straight line.
- R^2 (Square of the correlation coefficient)

$R^2 > 0.88$
Approved ✓

$$R^2 = \frac{[\sum(x_i - \bar{x})(y_i - \bar{y})]^2}{\sum(x_i - \bar{x})^2 \sum(y_i - \bar{y})^2}$$

By calculator

$R^2 < 0.88$
Rejected ✗

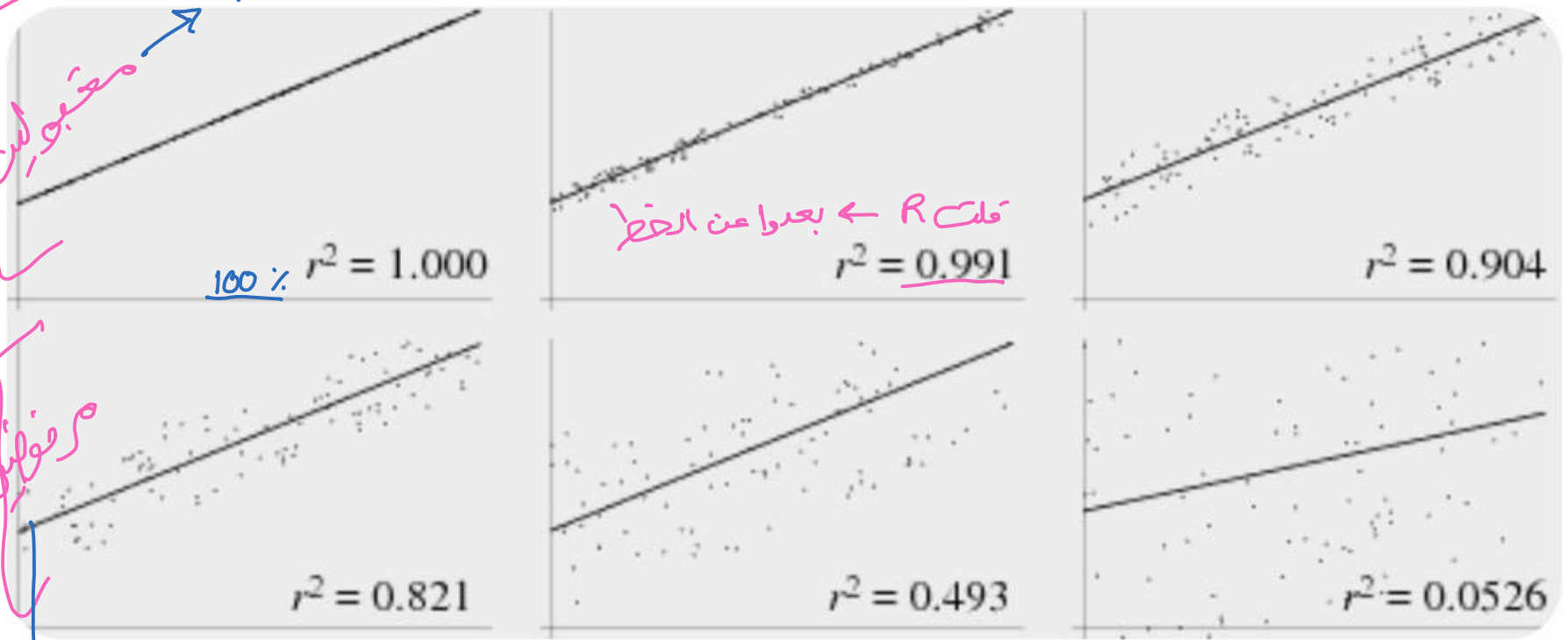
مجموع x
عدد

[X]	y → Response
1	
2	
5	
10	
25	
50	
100	

$\bar{y} = \frac{\text{مجموع y}}{\text{عدد}}$

معقولين $\rightarrow R^2 > 0.88$

غير معقولين $\rightarrow R^2 < 0.88$



INDETERMINATE OR RANDOM ERRORS

Standard deviation: $s = \sqrt{\frac{\sum_{i=0}^{i=N} (x_i - \bar{x})^2}{N-1}}$

Variance: s^2

No
of x

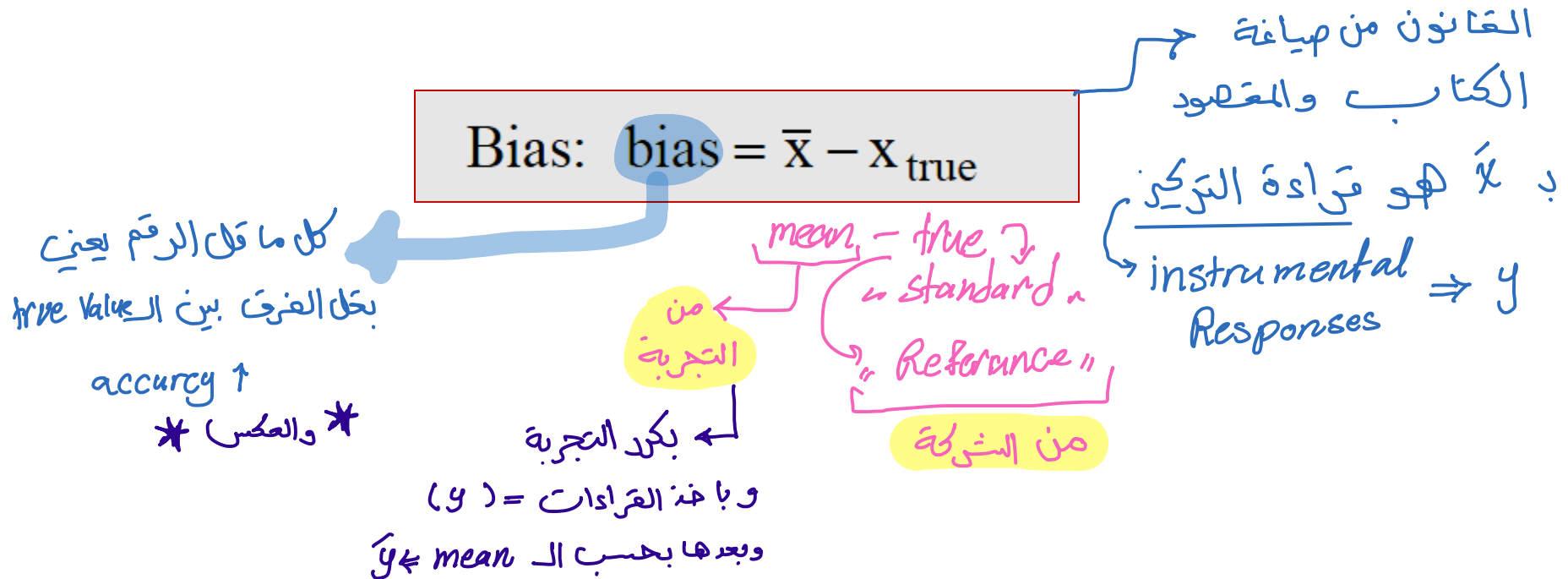
Relative standard deviation: $RSD = \frac{s}{\bar{x}}$

نسبة
الخطأ
المتوسط

Standard deviation of mean: $s_m = \frac{s}{\sqrt{N}}$

FIGURES OF MERIT: ACCURACY

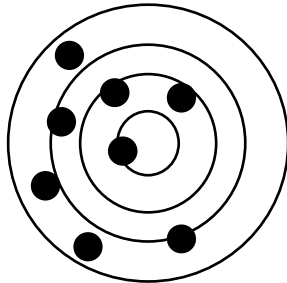
DETERMINATE ERRORS (operator, method, instrumental)



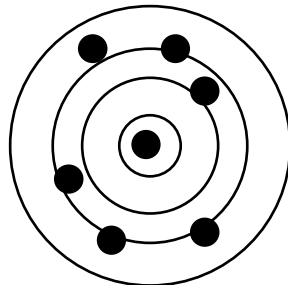
Illustrating the difference between “**accuracy**” and “**precision**”

بجود صحتهم الى
مما لا

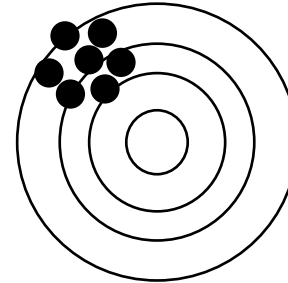
بجودهم
عن بعض



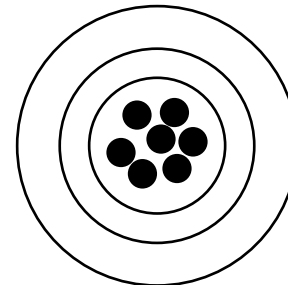
Low accuracy, low precision



High accuracy, low precision

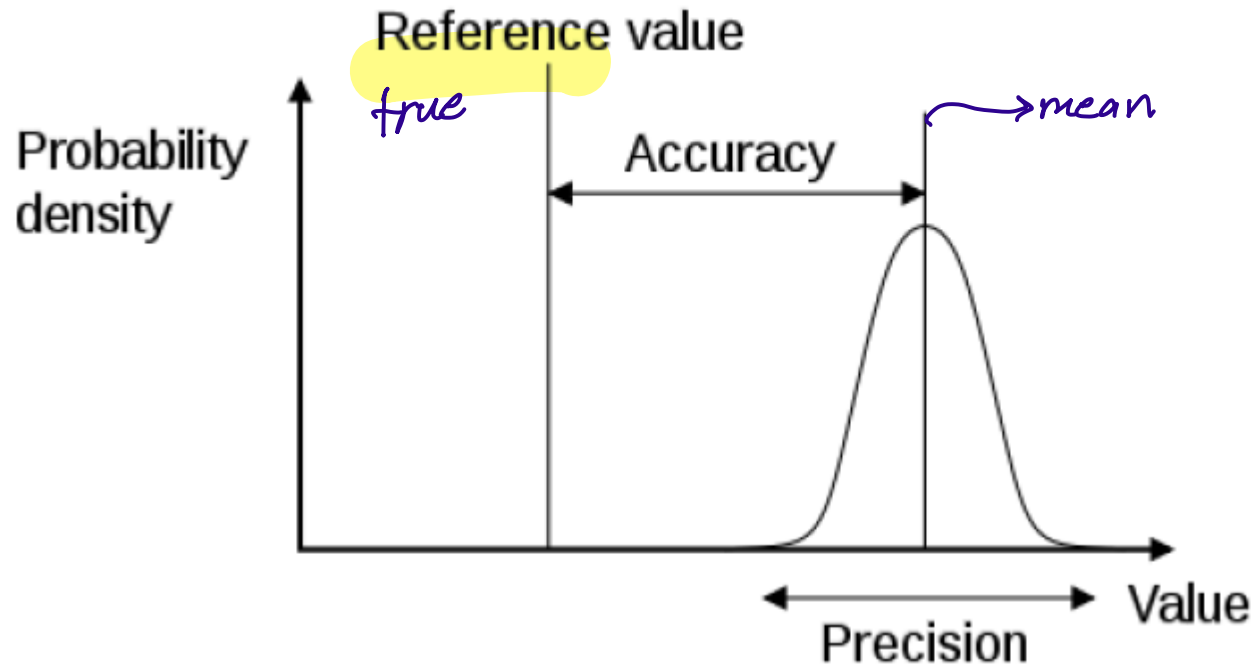
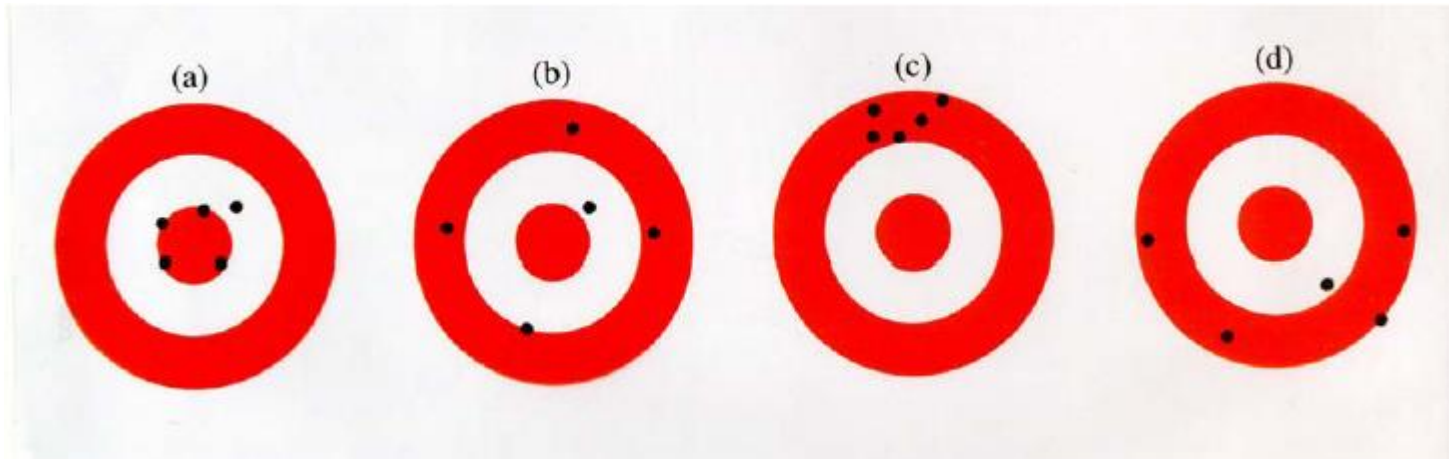


Low accuracy, high precision



High accuracy, high precision

Accuracy vs Precision



FIGURES OF MERIT: SENSITIVITY

Method → slope
Method at limit con = $\frac{\mu}{S_s}$

Indicates the response of the instrument to changes in analyte concentration or a measure of a method's ability to distinguish between small differences in concentration in different samples.

Effected by slope of calibration curve.

$$\text{Sensitivity} = \gamma = m/S_s$$

m = slope;

S_s is the standard deviation of measurement

(larger slope of calibration curve m, more sensitive measurement)

قبل الـ LOD, LOQ, 10% في ← 3 Definitions

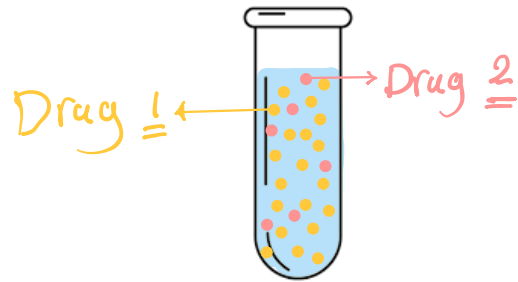
Sample

اخذت من

• مريض (Blood, Urine)

• مصدر آخر (نخر, بركة, Rat Plasma,)

و مطلوب مني اعرف وين المشكلة, شو الملوث
ونسب الأدوية او الملوثات



Blank

فاخنة ⇒

* لو Sample اللي مطلوب

احللتها عبارة عن Urine

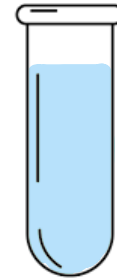
← بجيب Urine لشخص

سليم - ما بياخذ دواء

* لو العينة من نخر ملوث

← بجيب Pure H₂O

يعني Blank من نفس نوعي الـ Sample



Blank

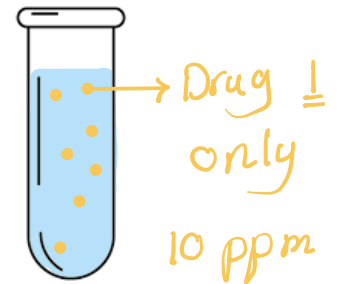
Pure H₂O

Stock

فيها ترا كين

مدرسة وانا

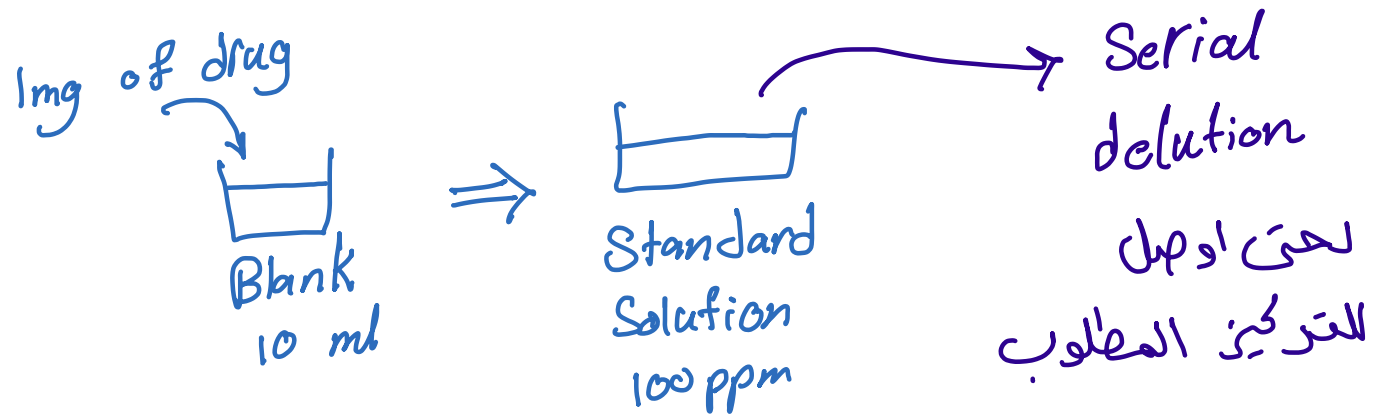
اللي بتحكم فيها



بدین عمل تحلیل Urine امریست ما خذ دوا و عمله مست حل
Sample → drug in urine

Blank → Pure urine without the drug ← شحلی طبیعی

Blank + محلوله → Standard Solution



FIGURES OF MERIT: LIMIT of DETECTION (LOD)

أقل تركيز يمكن للتحليل
قياسه بدقة
من دقة عالية

Signal must be bigger than random noise of blank

Minimum signal: $\text{Signal}_{\min} = \text{Av. Signal}_{\text{blank}} + k s_{\text{blank}}$

From statistics $k=3$ or more (at 95% confidence level)

Typically 3 times the signal-to-noise
(based on standard deviation of the
noise)

or

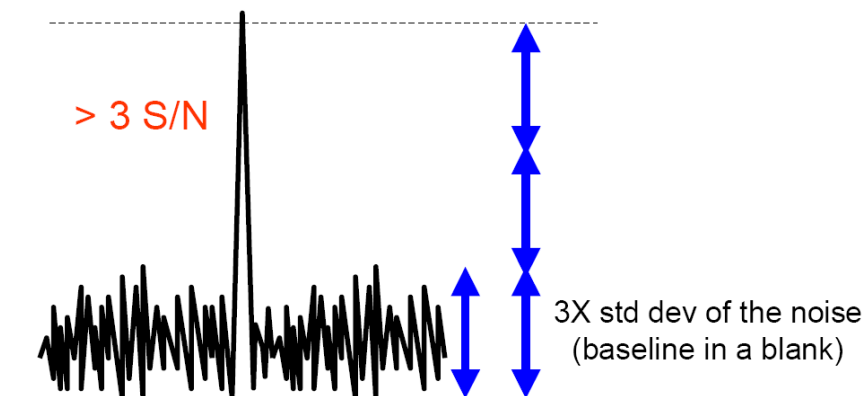
$\text{LOD} = 3 \times \text{Standard deviation of blank} /$
Slope

$\text{LOD} = 3 \times S_b / m \rightarrow \text{slope}$

فقط
urine

هاد المستخرج

Is this peak real?



FIGURES OF MERIT: DYNAMIC RANGE

أقل تركيز يمكن ملاحظته
بسهولة

At limit of detection we can say confidently analyte is present but cannot perform reliable quantitation

LOQ (limit of quantification): [lowest] at which quantitative measurements can reliably be made. Equal to 10 x Average Signal for blank *i.e.* 10S_{bl}. or

$$\text{LOQ} = 10 \times \text{Standard deviation of blank} / \text{Slope}$$

$$\text{LOQ} = 10 \times S_b / m$$

Limit of linearity (LOL): when signal is no longer proportional to concentration

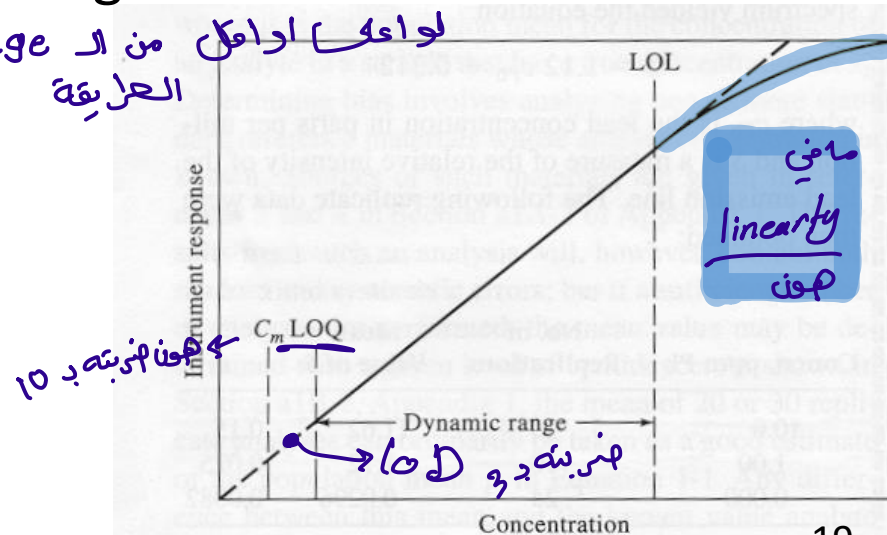
Dynamic range: the maximum range over which an accurate measurement can be made

From limit of quantitation to limit of linearity

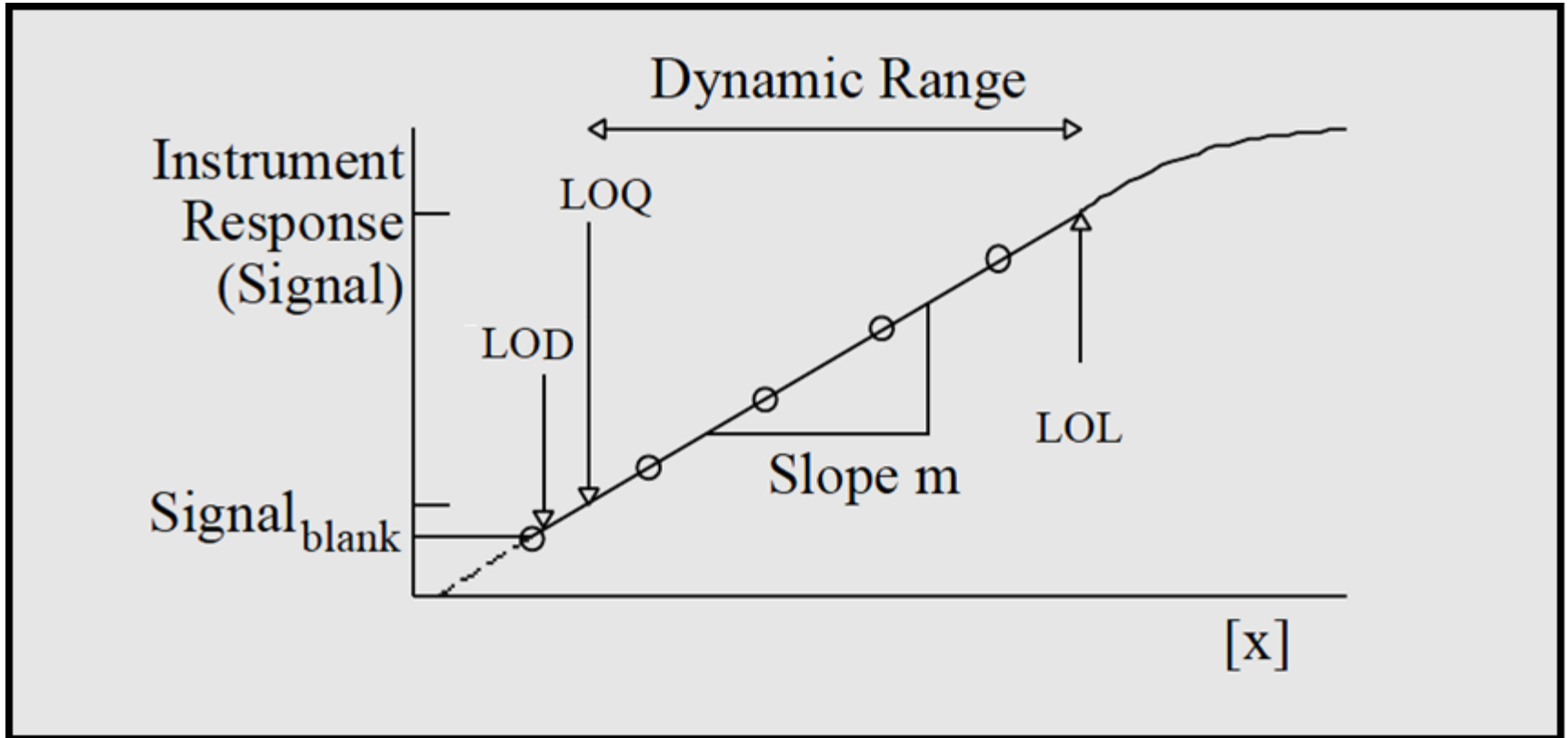
LOQ: 10 s of blank

LOL: 5% deviation from linear

$$\text{Dynamic range: } \frac{\text{LOL}}{\text{LOQ}} \quad 10^2 \text{ to } > 10^6$$



CALIBRATION CURVES (WORKING or ANALYTICAL)



غیرت ی قیب السلایدات
مشان معنی مع تلسن الشرح

CALIBRATION METHODS

Basis of *quantitative* analysis is magnitude of *measured property* is proportional to *concentration* of analyte

$$y = \text{Slope } x \pm \text{intercept}$$

اول اشي بعله بطلع ال calibration eqv.

$$\text{Signal} \propto [x] \quad \text{or} \quad \text{Signal} = m[x] + \text{Signal}_{\text{blank}}$$

$$[x] = \frac{\text{Signal} - \text{Signal}_{\text{blank}}}{m}$$

جزء المختبر العملي



بعمل اكثر من solution و بدخلهم على الجهاز
وياخذ القراءه

SAMPLE PROBLEM:

اول اسٹی لازم ایلج لہ eqn ← $intercept$ ← m بدی

Analyte Concentration (ppm*)	Absorbance
0.0 (blank) ہائی اسٹی	0.05
0.9	0.15
2.0	0.24
3.1	0.33
4.1	0.42

$$1 - \text{slope} = \frac{S_{xy}}{S_{xx}}$$

Define Variance and Covariance:

$$S_{xx} = \frac{\sum (x_i - \bar{x})^2}{N - 1} \quad S_{xy} = \frac{\sum (x_i - \bar{x})(y_i - \bar{y})}{N - 1}$$

$$\bar{x} = 2.02 \quad \bar{y} = 0.238$$

$$S_{xx} = \frac{(2.02^2 + 1.12^2 + 0.02^2 + 1.08^2 + 2.08^2)}{4} = \frac{10.828}{4} = 2.707$$

$$S_{xy} = \frac{(-2.02 \times -0.188) + (-1.12 \times -0.088) + (-0.02 \times 0.002) + \dots}{4}$$

$$= \frac{0.9562}{4} = 0.23905$$

$$2- \text{intercept} = \bar{y} - m\bar{x}$$

$$\text{Slope: } m = \frac{S_{xy}}{S_x} = \frac{0.23905}{2.707} = 0.0883$$

$$b = \bar{y} - m\bar{x}$$

$$\text{Intercept: } = 0.238 - (0.0883 \times 2.02)$$

$$= 0.0596 \rightarrow \text{لوكانت صابة}$$

Calibration expression is بزلج صابة

$$\text{Absorbance} = 0.0883[\text{Analyte (ppm)}] + 0.0596$$

$$y = 0.088x + 0.0596$$

هيب شو بستيقي من
المعادلة؟

- بجيب ال Sample وبعفها
بالجهاز وبتطلع قراءة y

25 ومن المعادلة بوجد $x \leftarrow \text{conc.}$

Example:

An analysis of the calibration data for the determination of lead based upon its flame emission spectrum yielded an equation: $S = 1.12C + 0.312$ where C is the Pb concn in ppm and S is a measure of the relative emission intensity. The following replicate data were obtained:

$$y = 1.12x + 0.312 \quad b = 0.312 \\ m = 1.12$$

Concn (C), ppm	No. of replicates	Mean value of S	Std. dev.
10.0	10	11.62	0.15
1.00	10	1.12	0.025
0.000	24	0.0296	0.0082

Calculate (a) the calibration sensitivity, (b) the analytical sensitivity at 1 and 10 ppm of Pb, (c) the limit of detection limit (LOD), and (d) the limit of quantification limit (LOQ)

Example Solution:

a) calibration sensitivity is the slope of the calibration curve = 1.12

b) Using $\gamma = m/S_s$: at 10 ppm $\gamma = 1.12 / 0.15 = 7.5$ at 1 ppm

$\gamma = 1.12 / 0.025 = 45$

c) $LOD = 3 \times S_b / \text{slope} = 3 \times 0.0082 / 1.12 = 0.0219$

d) $LOQ = 10 \times S_b / \text{slope} = 10 \times 0.0082 / 1.12 = 0.0732$