Cycle 2

Multistep synthesis of Phenytoin

- Phenytoin, sold under the brand name Dilantin among others.
- It is an anti-seizure medication.
- It is useful for the prevention of tonic-clonic seizures and partial seizures.
- Phenytoin blocks the spread of seizure activity in the brain by causing voltage dependent block of the voltage gated Na channels.
- It may also be used for certain heart arrhythmias or neuropathic pain.
- · It doesn't have sedative hypnotic activity or very little
- It can be taken intravenously or by mouth

General scheme of synthesis 2 aldehyde links with covalent bond vit BI (cataly SIt) Thiamine Hydrochloride NaOH Ethanol Benzoin Covalently H20 + NaOH/Ethanol Benzil 5,5-diphenylhydantoin (Dilantin) Cyclization Phenytoin can synthesize from benzaldehyde in three synthetic steps: Part 1: Benzoin condensation. Part 2: Oxidation of benzoin to benzil. Part 3: Condensation of benzil with urea to form phenytoin. You will see the mechanism for each steps.

Try to enjoy it

Experiment 2

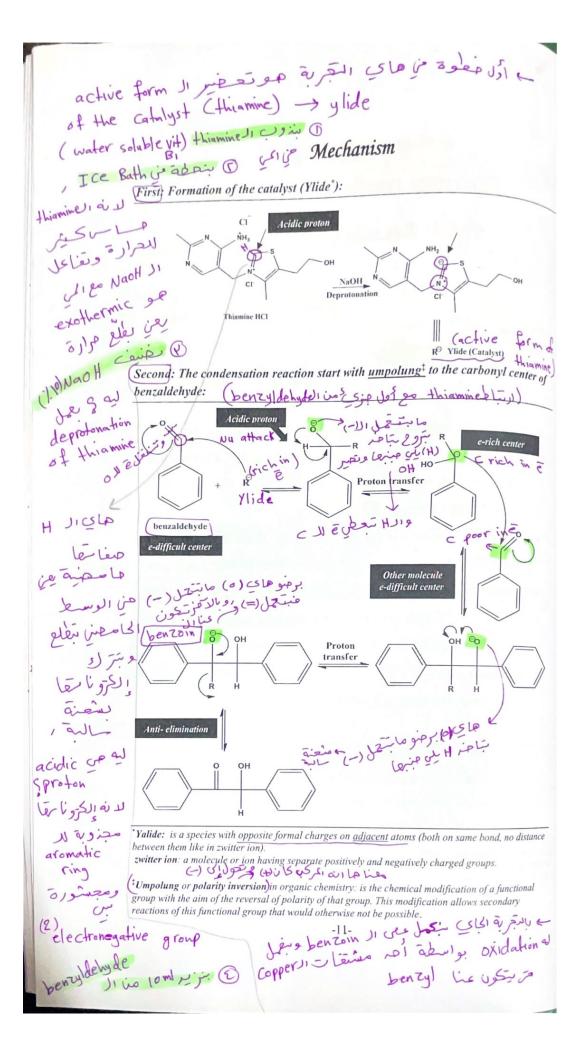
Multistep synthesis of Phenytoin

Part1: Benzoin Condensation

- Benzoin is produced by <u>dimerization of 2 molecules of benzaldehyde</u>, The benzoin condensation[§] is in fact a dimerization and not a condensation because a small molecule like water is not released in this reaction
- This reaction doesn't occur spontaneously. It needs a catalyst to render the aldehydic <u>carbon acidic</u> so that it can be deprotonated and function as a nucleophile to attack the second benzaldehyde carbon.
- Cyanide* $[K^+C \equiv N^-]$ can use as a catalytic reagent use <u>at 75°C</u> (give <u>faster rate</u>) but they are extremely poisonous.
- Thiamine (contains a thiazole unit) is the catalyst was used here at Room temp, (give very slow rate).
 - Thiamine (Vit B1) is non-toxic (edible) material
 - Thiamine is <u>heat sensitive</u> and may decompose if heated vigorously
 - Instead of running the reaction at elevated temperature. We will allow
 it to proceed closer to room temperature for at least 24 hours

Refer to appendix VIII: Benzoin condensation using cyanide.

^{*}Condensation reaction: a reaction in which two molecules combine to form a larger molecule, producing a small molecule such as H₂O as a by-product.



Procedure

- 1. In a stoppered E. Flask, prepare a solution of 1.04 g of thiamine hydrochloride in 3 ml of water[‡].
- When all the thiamine hydrochloride has dissolved, add 8 ml of 95% ethanol with mixing.
- 3. Cool the solution for a few minutes in an ice bath [because thiamine is heat sensitive].
- 4. Very carefully and slowly add 3 ml of 10% NaOH with mixing "making sure that the temperature of the solution never rises above 20 C" [because NaOH with water exothermic].
- 5. Add to the mixture 7 ml of pure Benzaldehyde with mixing.
- 6. Stopper the flask and allow it to stand at room temp at least overnight (longer period do no harm). At the end of the reaction period, the benzoin should have separated as fine crystals.
- 7. Cool the reaction mixture in an ice bath to complete the crystallization. اعدا عدا عدا المادة ال
- 8. Collect the product by vacuum filtration and wash them thoroughly with two 7 ml portions of cold 50% ethanol and several portions of water [color change from yellow to white].
- 9. Drain well and leave your product to dry (no need for recrystallization*).

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	Weight (gm)	M.P	تكون
Benzoin			fine he
2 P :			crystals

3 Brain storming Question

3.1 Discus the chirality of the product, does it optically active? Explain your answer.

^{*}Water is required in this reaction BUT in specific amount:

Too much water will force benzaldehyde out of the solution preventing an efficient reaction

⁻ Too little water prevents thiamine-HCl from dissolving.

Re-crystallizion an done using 95% ethanol.

[§] Ethanol to remove excess benzaldehyde and water to get rid of thiamine.

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Sara Jammain



Artery Academy