* ADTO releas Thist center

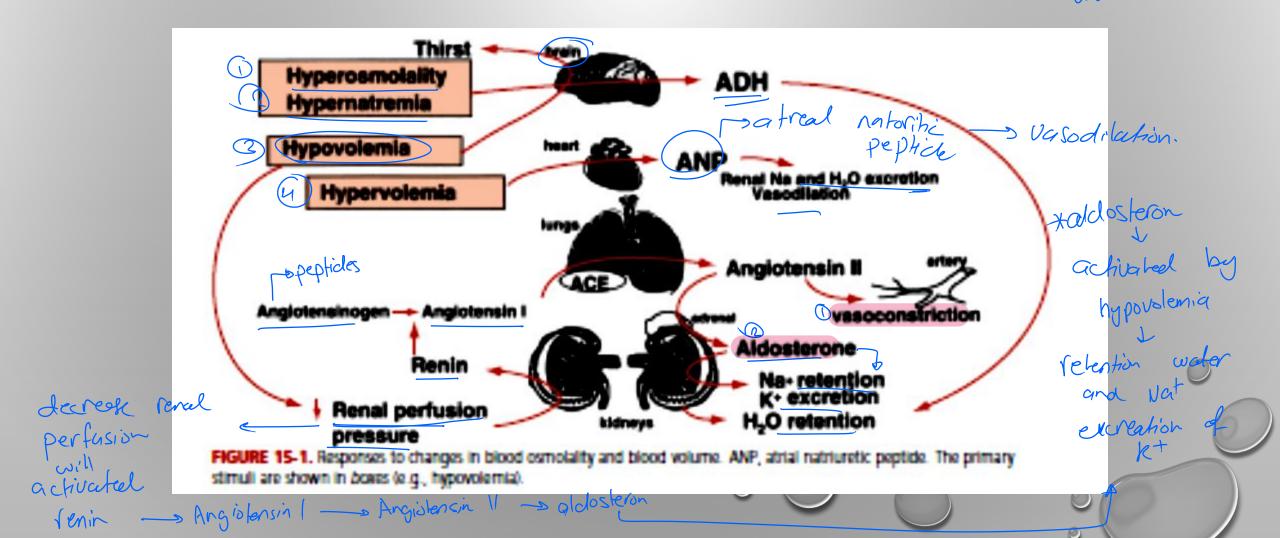
- 1. hypernathemia 2. hyperosmolarity Lo 3. Hypovdemia

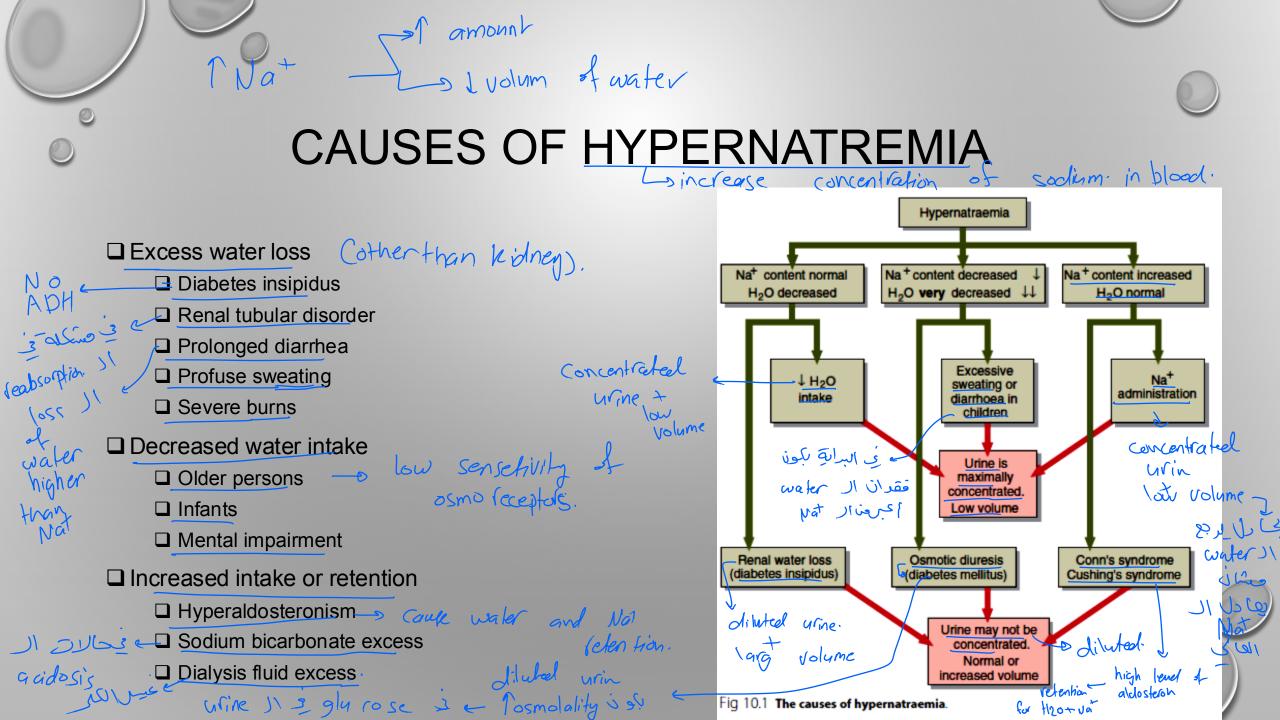
-orcab sorphon water.

تفريغ المحاضرة الرابعة اللهم لا سهل إلا ما جعلته سهلا 🎔 🕰

REGULATION OF SODIUM

* AND oHypervolenia excreation + Nat water + Nat





HYPERNATREMIA (150 MMOL/L) RELATED TO URINE OSMOLALITY

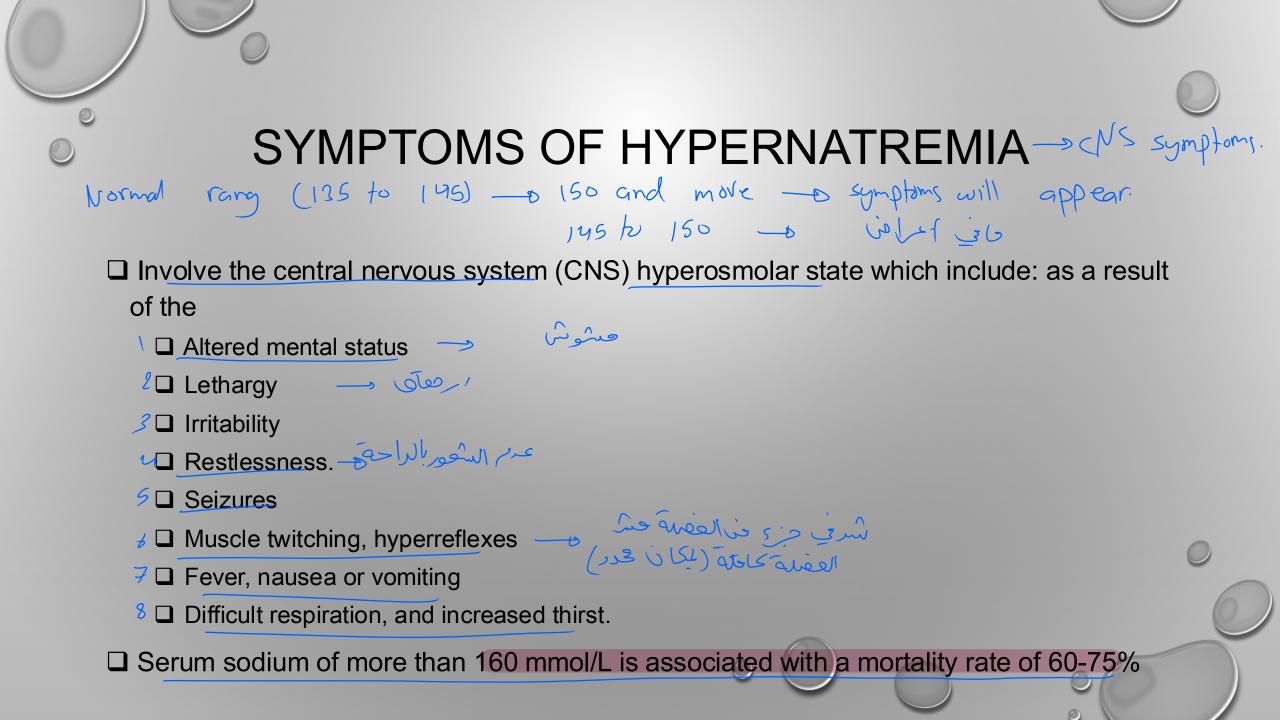
- (diluted, unine) ☐ Urine osmolality <300 mosm/kg
 - Diabetes insipidus (impaired secretion of ADH respond to ADH) or kidneys cannot
- ☐ Urine osmolality 300-700 mosm/kg
 - Partial defect in ADH release or response to ADH
- Osmotic diuresis (diabets mellitus / high Calicum level word mine)

 Urine osmolality > 700 mosm/kg (concentrated urine)

 How loss.

Loss of thirst

- ☐ Insensible loss of water (breathing, skin) / ☐
- □ GI loss of hypotonic fluid (vomiting, diamia
- ☐ Excess intake of sodium





- ☐ Treatment is directed at correction of the underlying condition that caused the water depletion or sodium retention.

 □ Treatment is directed at correction of the underlying condition that caused the water depletion or sodium retention.
 - Me vin Let vine lue.
- ☐ The speed of correction depends on the rate with which the condition developed
- ☐ Hypernatremia must be corrected gradually because too rapid a correction of serious hypernatremia (>160 mmol/L) can induce cerebral edema and death.

The maximal rate should be 0.5 mmol/L per hour

18 15 315 rate 11 131 d



HYPONATREMIA

Lo low concentration of Sodium in blood.

☐ Hyponatremia is defined as a serum or plasma level <135 mmol/L.

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☐ Levels below 130 mmol/l are clinically significant.

☐ Hyponatremia can be assessed by the cause for the decrease or with

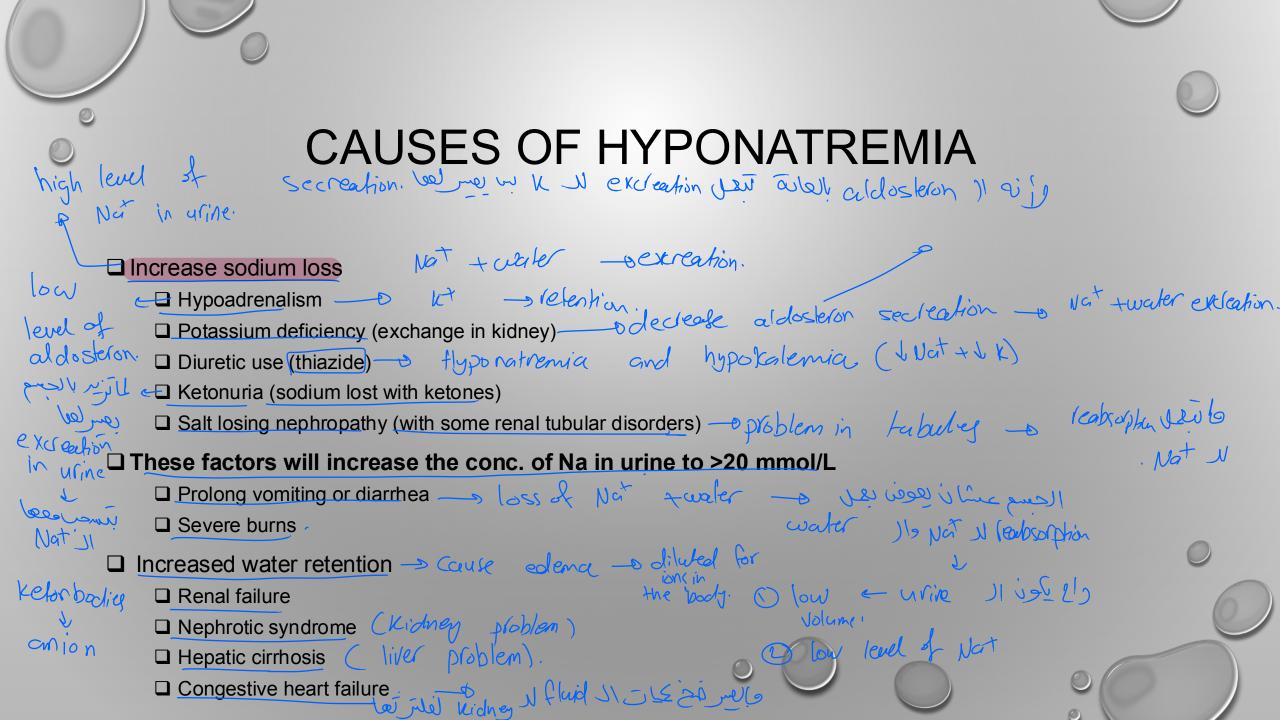
ل لذم اربي سو

the osmolality level.

level Nat Joi un vissel de

+ we can measured the concentration in serum, and plasma or win







CAUSES OF HYPONATREMIA

Water imbalance

Excess water intake (polydipsia, increased thirst): may cause mild or severe hyponatremia if water intake was chronic. In a normal individual, excess intake will not affect Na levels.

SIADH causes an increase in water retention because of increased ADH produc

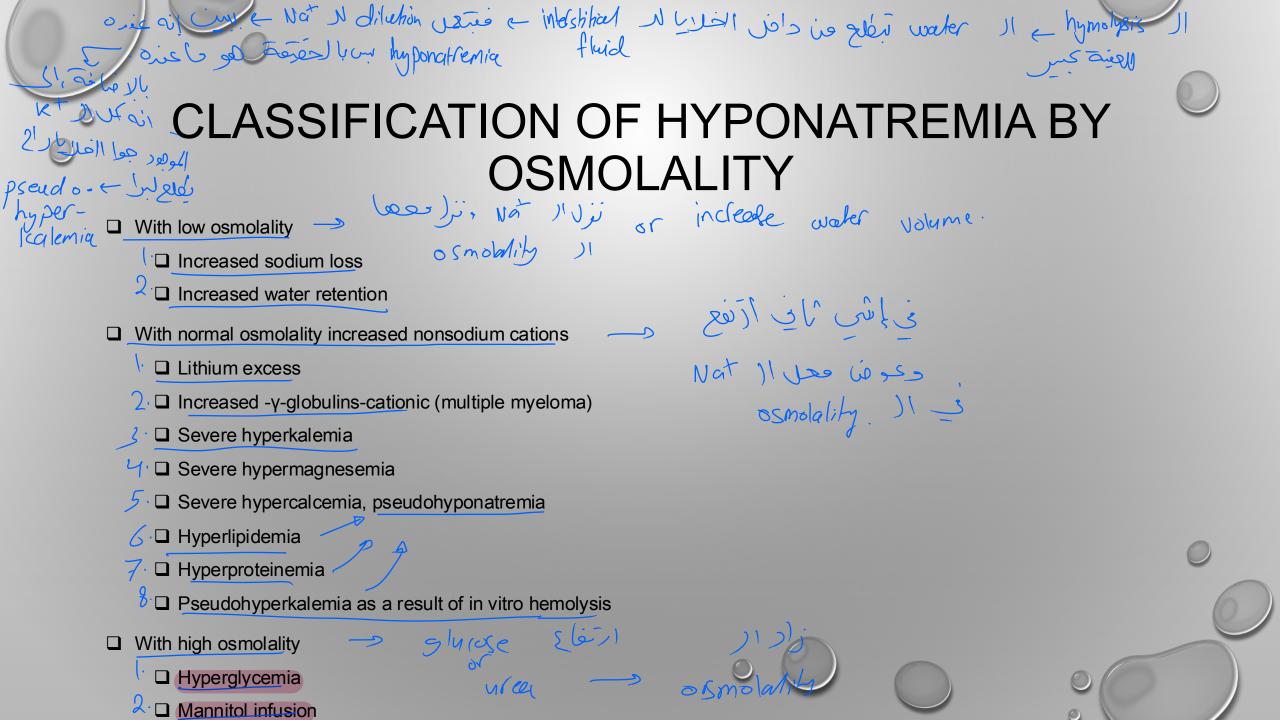
□ SIADH causes an increase in water retention because of increased ADH production which is associated with pulmonary disease, malignancies, CNS disorders, infections.

Pseudohyponatremia by measuring Na using indirect ISE (which dilutes sample prior to analysis),

in a patient with hyperproteinemia or hyperlipidemia.

3) hymolykis of sample

hypo nationia



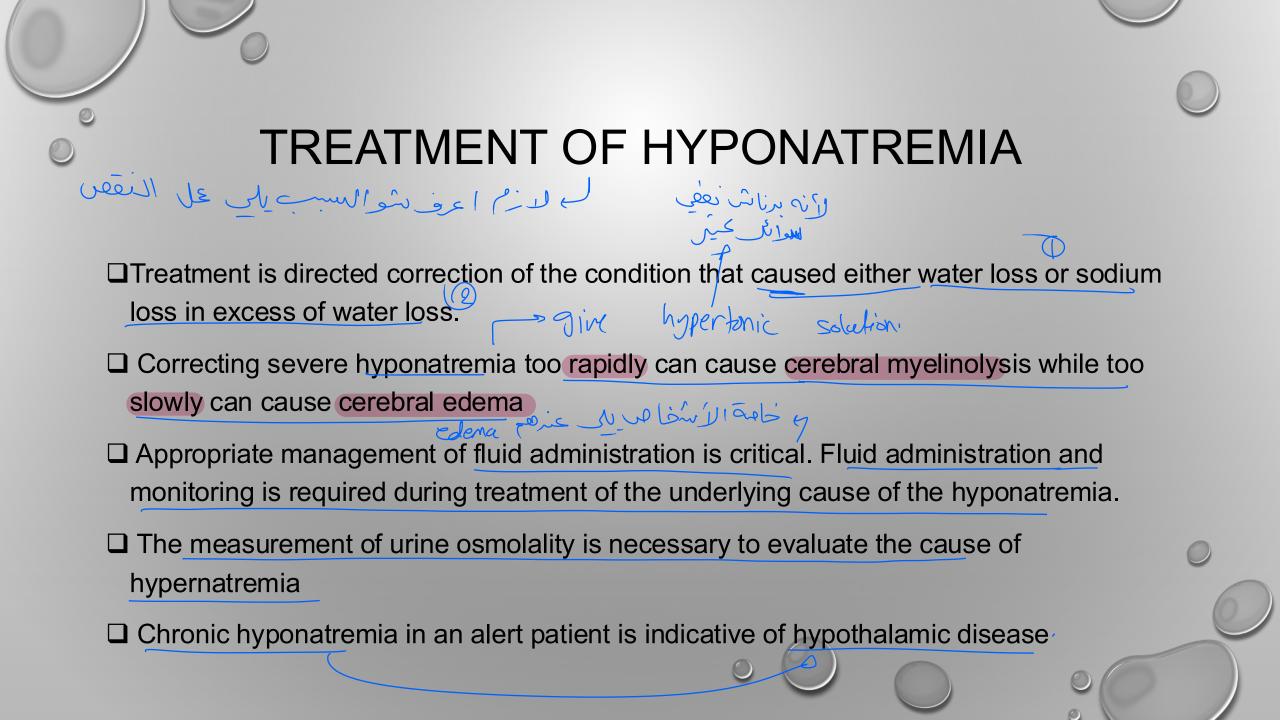


- □Symptoms depend on the serum level.
 - Between 125 and 130 mmol/l: symptoms are gastrointestinal (Mausea and Comiting)

□Below 125 mmol/l: more severe neuropsychiatric seen including nausea and vomiting, muscular weakness, headache, lethargy, and ataxia.

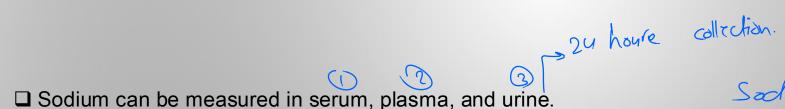
☐ More severe symptoms also include seizures, coma, and respiratory depression

death





DETERMINATION OF SODIUM



Sadinm. le s finite (vij)

- ☐ When plasma is used, lithium heparin, ammonium heparin, and lithium oxalate are suitable anticoagulants.
- Hemolysis does not cause significant change in serum or plasma values as a result of decreased levels of intracellular sodium. however, with marked hemolysis, levels may be decreased as a result of a dilutional effect
- ☐ Whole blood samples may be used with some analyzers.
- ☐ The specimen of choice in urine sodium analyses is a 24-hour collection.
- ☐ Sweat is also suitable for analysis.

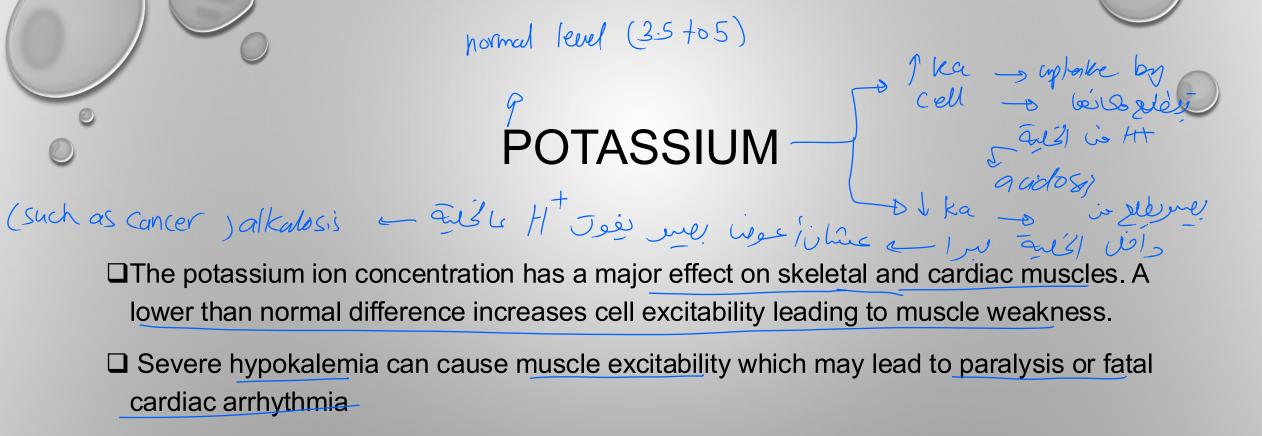


POTASSIUM - omajor intracellular cation.

- ☐ Potassium is the major intracellular cation in the body with a concentration 20 times greater inside the cells than outside
- ☐ Many cellular functions requires that the body maintains a low ECF concentration of K. As a result, only 2% of the body's total potassium circulates in the plasma
- ☐ Function of potassium in the body include
 - ☐ Neuromuscular excitability
 - 2 Contraction of the heart

high level of K+ ECF - Suis 11 get and 224 mes 11 acidosis. Ver ale amin 131 e ECF U Ht level 224 Pole Threes 4- Hydrogen ion concentration

3 TICF volume (intracellular fluid volume) - prevent swelling or shrinking of



□ Hypokalemia decreases cell excitability resulting in an arrhythmia or paralysis

0

☐ the heart may cease to contract in extreme case of hypokalemia or hyperkalemia

Potassium concentration affects hydrogen ion concentration in the blood. In hypokalemia, when potassium ion is lost from the blood, sodium and hydrogen ions move to into the cells. The hydrogen ion concentration decreases in ECF resulting to alkalosis

