

- Body composition

در حالت عادی

Water

أهم الأ
Electrolytes
موازنة على جانبي
membrane
خارجي وتحتلهم

The concentration of ions inside the cells and in plasma is ☐
maintained by passive diffusion and active transport through
ATPase-dependent ion pump

Most biological membranes are permeable to water but not ☐
ions

كيف يخرج الماء
Water and sodium output ☐
Kidneys and gastrointestinal tract ☐
Sweat and expired air: about 1L daily ☐

Factors that affect the flow of water across the membrane ☐
الأثر
يحلل عوامل الضغط
Ions and proteins at one side of the membrane ☐
Blood pressure ☐



Clinical features of hydration problems

حالات مشاكل المياه

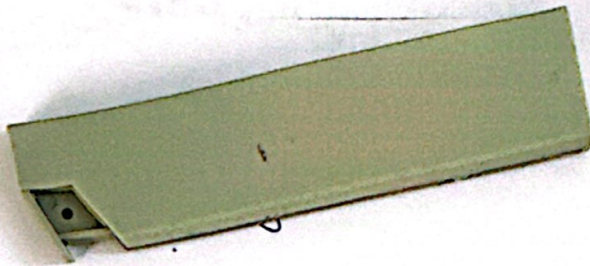
Table 6.1 The principal clinical features of severe hydration disorders

Feature	Dehydration	Overhydration
Pulse	↑ Increased	Normal
Blood pressure	↓ Decreased	Normal or increased
Skin turgor	↓ Decreased	Increased
Eyeballs	Soft/sunken	Normal
Mucous membranes	Dry	Normal
Urine output	↓ Decreased	May be normal or decreased
Consciousness	↓ Decreased	↓ Decreased

طراوة الجلد
إذا العين
غائبة أو
dry

بمتفاني
التوسع
مسنزاد
بالسوائل

إذا شفي شرب كثير يبلع بالبول
مضطرب يكون فائدة مشاكل في الكلى
Kidney



Control of water balance

Sensoria - hypothalamic osmoreceptor = حساسية osmo موجودة بالرغاي مسؤولة

Both intake and loss of water are controlled by osmotic gradient across cell membrane in the brain hypothalamic osmoreceptor centre

These centres control thirst and secretion of antidiuretic hormone (ADH)=AVP (arginine vasopressin hormone)

Thirst is the major defense mechanism against hyperosmolality and hypernatremia

Antidiuretic hormone:

Is polypeptide with t. of 20 min

Synthesized by the hypothalamus and secreted by the posterior pituitary

2% increase in osmolality lead to 4 times increase in ADH

Low blood pressure and severe hypovolemia stimulate ADH release

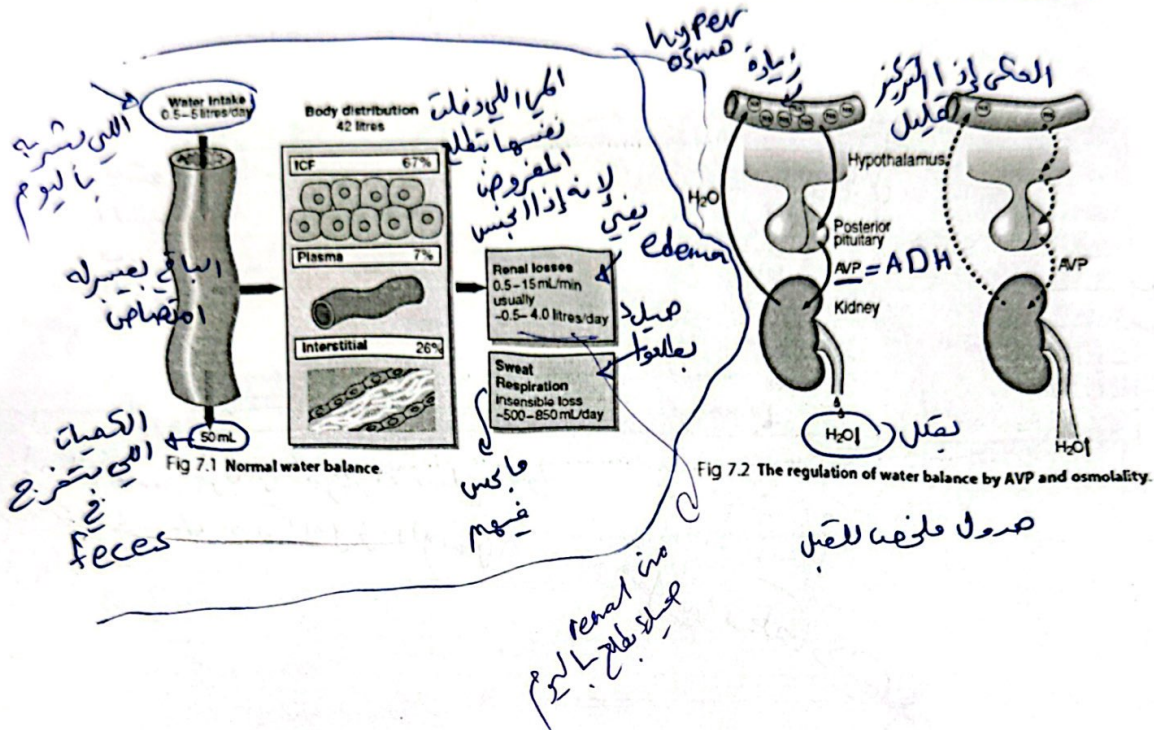
Stress due to vomiting, nausea and pain may increase ADH secretion

ADH act by increasing the reabsorption of water in cortical and medullary collecting tubules

هناك كمية الماء الموجودة
بالجسم
يرجع أو يقل
هناك حساسية
(لا تكفي العطش)



Control of water balance



* Control of water balance *

- نادراً ما يحدث نقص الماء في الشخص العادي
- Hypernatremia rarely occurs in a person with a normal thirst mechanism and access to water, it becomes a concern in:
- Infants
 - Unconscious patients
 - Anyone who is unable to drink or ask for water.
 - People who are older than 60 where osmotic stimulation of thirst progressively diminished
 - In the older patient with illness and diminished mental status, dehydration becomes increasingly likely example of the effectiveness of thirst in preventing dehydration
 - A patient with diabetes insipidus (no ADH) may excrete 10 L of urine daily, but as water intake matches output, plasma sodium remains normal
- مشكلة في الغدة النخامية (Pituitary gland)

لأنه في ADH
إذا شرب معهم ما في ماء أو ما إذا شرب مع hypernatremia

osmolality الضغط الأسمولي

Physical property based on the conc. of solutes (in mmol) per kg of solvent (w/w). This affects different properties of solution as:

- Freezing point depression
- Vapor pressure decrease

Increase in osmolality will induce secretion of ADH enzyme while decrease in osmolality will lead to turning off ADH secretion

Osmolal gap is the difference between the measured osmolality and the calculated osmolality

Osmolal gap indirectly indicates the presence of osmotically active substances other than sodium, urea, or glucose such as ethanol, methanol, ethylene glycol, lactate, or β -hydroxybutyrate.

هناك فرق بين الضغط الأسمولي والضغط الأسمولي

* هناك فرق بين الضغط الأسمولي والضغط الأسمولي
(measured vs calculated osmolality)
(↓ vapor pressure, ↓ freezing point)

Significance of osmolality

Because it is the parameter by which the hypothalamus responds

It affects Na concentration as it represents 90% of osmotic activity in Plasma

Na concentration is also affected by blood volume

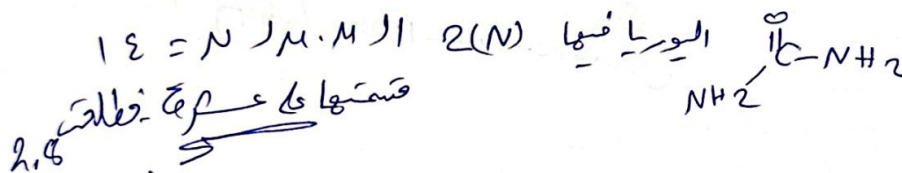
Determination of osmolality

Osmolality may be measured in serum or urine. ☐

Plasma use is not recommended because osmotically active substances may be introduced into the specimen from the anticoagulant. ☐

Samples must be free of particulate matter to obtain accurate results. ☐
 هذا يعني إزالة كل شيء عالقة في العينة turbidity

Turbid serum and urine samples should be centrifuged before analysis. ☐
 هذا يعني



Determination of osmolality

Osmometers are standardized by NaCl solution, then the freezing point of the sample is measured and this is compared to the calculated value as double of serum sodium or according to the following 2 formulas: ☐

$$2 \text{ Na} + \frac{\text{glucose (mg/dL)}}{20} + \frac{\text{BUN (mg/dL)}}{3}$$

49 g/dL
 40 g/L

هذا على كل شيء
 لهذا

$$1.86 \text{ Na} + \frac{\text{glucose}}{18} + \frac{\text{BUN}}{2.8} + 9 \quad (\text{Eq 15-1})$$

$$1.86 \text{ Na} + \frac{\text{glucose}}{18} + \frac{\text{BUN}}{2.8} + 9$$

هذا يعني ان كل شيء
 في المول

$$\frac{\text{mg} \times 100 \text{ mmol/L}}{\text{dL}} \quad \frac{\text{mmol/L}}{\text{mg}} \quad \frac{\text{mmol}}{\text{mg}}$$

Mwf mg/mmol

measured osm... = 290 mOsm/kg (by lab) \therefore is body
 $\text{Na}^+ = 142 \text{ mmol/L}$ / Glucose = 90 mg/dL / BUN = 14 mg/dL
 $1.86 \times 142 + \frac{90}{18} + \frac{14}{2.8} = \text{calculated osm}$
 $= 264 + 5 + 5 = 274 \text{ mOsm/kg}$ (Normal ranges)
 $= 283.1 \text{ mOsm/kg}$

= osmolal gap

$290 - 283.1$

= 6.9

reference

TABLE 15-1 REFERENCE RANGES FOR OSMOLALITY

Serum	275-295 mOsm/kg
Urine (24-h)	300-900 mOsm/kg
Urine/serum ratio	1.0-3.0
Random urine	50-1200 mOsm/kg
Osmolal gap	5-10 mOsm/kg

در این حالت
 جفاف

در این حالت
 ADH

جفاف نسبی
 لا زیم

Electrolytes, Sodium (Na)

Body contains about 3000 mmol of sodium mainly in ECF ☐

Extra cellular fl.

Sodium daily intake is about 60-150 mmol ☐

Sodium balance is regulated by blood flow and aldosterone ☐
 (hormone secreted by adrenal cortex)

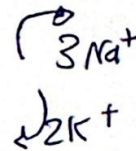
Electrolytes, Sodium (Na)

Sodium is the most abundant cation in the ECF (90% of all extracellular cations) and largely determines the osmolality of the plasma.

Sodium concentration in the ECF is much larger than inside the cells, because a small amount of sodium can diffuse through the cell membrane.

To prevent equilibrium from occurring, active transport systems, such as ATPase-dependent ion pumps (moves 3 Na out of cell for each 2 K moving into the cell) are present in all cells

أو hemolysis ممكن يا شغل زيادة
الصوديوم؟ لوذا بسيط ال hemo غلبا شتر
لوذا اكثير يا شتر لانه فرق ال تركيزه كبير.
ولا نه جعله سائله لانه بيطلع عليه / لها
صبيش Pseudo hyponatremia



Regulation of sodium

الصوديوم اللى بنوكله بيروح اقله ل renal

The plasma sodium concentration depends on: the intake and excretion of water and the renal regulation of sodium

Three processes are of primary importance:

- (1) The intake of water in response to thirst, as stimulated or suppressed
- (2) the excretion of water, largely affected by ADH release by plasma osmolality volume or osmolality in response to changes in either blood

- (3) the blood volume status, which affects sodium excretion through and ANP (atrial natriuretic peptide) aldosterone- angiotensin II

The kidneys have the ability to conserve or excrete large amounts of sodium, depending on the sodium content of the ECF and the blood volume, normally, 60-75% of filtered sodium is reabsorbed in the proximal tubule

some sodium is reabsorbed in the loop and distal tubules (controlled by aldosterone) exchanged for K in the connecting segment and cortical collecting tubule.

thirst
تلة الماء
2
distal tubule