# Clinical Biochemistry for Pharmacy

Introduction

Dr. Iman Mansi

طبعا كلنا عارفين إنه المادة بتشرحها دكتورة إيمان وكلنا مجربين امتحاناتها وموادها من قبل نصيحة لا تراكموا خصوصا انه الفصل كثير زخم وحاولوا تربطوا المعلومات لأنه الدكتورة كثير بتحب الربط وبالاخير المادة بتحكي عن شغلات مهمة كثير إلك كصيدلاني فلا تضيعوها من ايدكم وحاولوا تستفيدوا قدر المستطاع

Ref. An illustrated colour text in clinical biochemistry, Gaw, 4th ed, 2008

يعطيكم العافية ان شاء الله يكون فصل خير علينا حميعا الله

#### CLINICAL BIOCHEMISTRY

- Clinical biochemistry, chemical pathology and clinical chemistry its important for you as a phormacist for:

  1. help patient with diagnosis and progressis the disease.
  2. to limit the patient from self treatment.
- The test must be <u>performed</u> as accurately as <u>possible</u>
- Biochemical tests are used extensively in medicine for both diagnosis and prognosis of diseases
- Lo The test should be very accuratly to prevent the wrong diagnosis and treatment for disease. The prevent the wrong
  - Your role as a pharmacist (Terminology knowledge, understand basis of lab test, effect of disease and drugs on the diagnostic test)

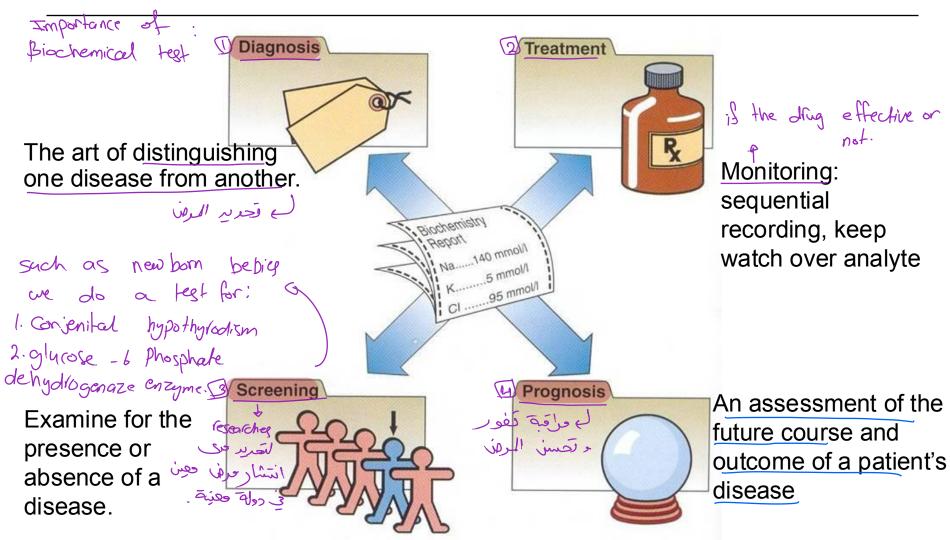
    on the diagnostic test)

    الم يمين في المالي الفعال الفعال المعالية المحالية المحالية

EX for understanded basic of lab test and effect of disease and drugs on the diagnostic test.
The measuring of glucose level in the blood:
الم المام ملا معنه المام عند المام
2. Analysis of Sample in the lab by using kit or by using glucometer - Sies 21, airell uls als una glucose level 11 551 à s siè
* in the Kit — low interferences.  * in the glucometer — there are interferences.
There are many drugs will incease the level of glucose in the blood
Such as Conticosteroids
م فياذا إحال مريفياعنده hypuglycemia منديفياع باخداً ودي
أولا لأنه عكن تكون الـ hypoglycemia العواء مشر خلا
E Il insulin clèb amas.
في عندي في عين يتم علهم الحريم الرافقال حديث الولادة:
1. Conjenter hypothyrodism - > === , 11 === les vez
2. glucose & phosphate dehydrogenase en agre.
إذا كان في عندي defect في الأنزليم هذا ب الففل راح بكون مصاب مرفن التفولي -

#### Biochemical tests

#### Qualitative and quantitative analysis



# The place of clinical biochemistry in medicine

Clinical biochemistry tests comprise over one-third of all hospital laboratory investigations

The patient they have colonic cancer

Lo the level of CFA (cancer Biomarker)

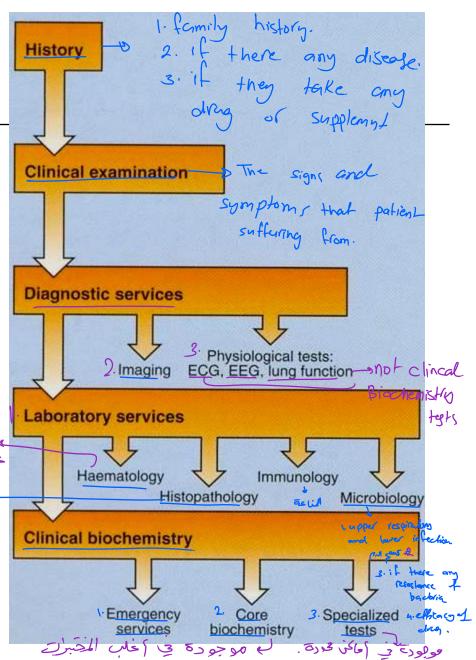
high.

The field of CFA (cancer Biomarker)

high.

The field of CFA (cancer Biomarker)

Also will also with a property of the colonians of malegnant.



## The clinical biochemistry tests

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Sodium, potassium, chloride and bicarbonate

Urea and creatining

id the patient have:

nausea, vomiting dicivia

(cause dehydiation).
Urea and creatinine -okidney belections to E + Go ply cose.
Calcium and phosphate - bons
Total protein and albumin - liver and kickey
Bilirubin and alkaline phosphatase - liver and galbladder
                                 high level if there a problem in bors.
ALT and AST -Liver
y-glutamyl transpeptidase - Liver and galbladder. - GGT
                              ورفع عندال خالعاله
Creatinine kinase — heart
H+, PCO2, and PO2 (blood gases) - Respiratoly Wast disease (asthma)
Glucose - Matchie patien, if patient take any drug will increase level of
Amylase - pancerage, GI
```

### The clinical biochemistry tests

#### Specialized tests

Hormones -> Sex hormons.

Specific proteins - ox- Keto protein

Trace elements -> zinc / Conium

Vitamins -> B12, vitamin D.

Drugs

Lipids and lipoproteins

LDL, ADL or **DNA** analysis cholesterol.

#### 3. Emergency tests

Urea and electrolytes

Blood gases

Amylase - Pancerialites.

Glucose - hypo or hyperglycemia.

<u>Salicylates</u> —

**Paracetamol** 

a cysteic Glorosis. Calcium

Generycine -scrious

family disease while blood cells

#### Importance of clinical chemistry

- A working knowledge of clinical chemistry and biochemistry is essential for the pharmacist to effectively communicate and interact with other health professionals.
- This course provides the basic information in lab medicine that is necessary for the pharmacist.

# Sample collection and processing

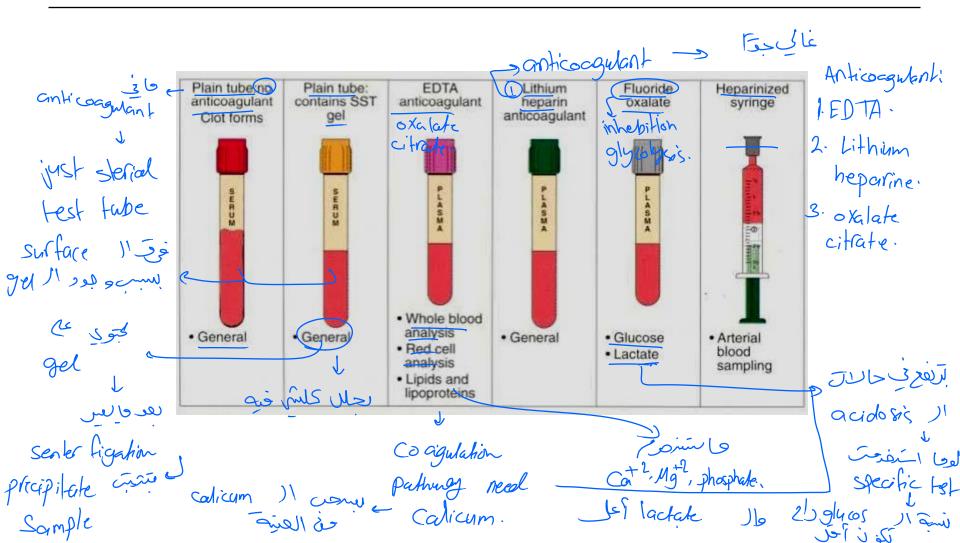
- Specimens used for biochemical analysis:
- Venous blood, arterial blood, capillary blood
- serum vs plasma specalical
- Urine tube without anything of tube contains coagulation factor.

  Faeces currous tract infection, keton for total protein, test.

  -CSF bleedly, infection
- Sputum and saliva -> upper respiratory tract infection. resistance.
- Tissue and cells 6.
- Aspirates, e.g. pleural fluid, synovial fluid, intestinal, pancreatic pseudocytes
  - Calculi (stones)

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# Types of containers used in the collection of blood specimens

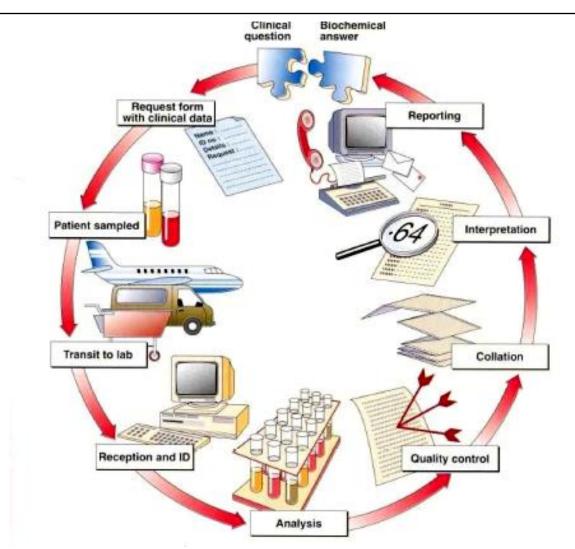


هیك بتكون خلصت اول محاضرة ان شاء الله إنها تكون بسيطة ويعطيكم العافية ورميلتكم مرام الزيادى  $\ref{eq}$ 

#### Sampling errors

- Blood sampling technique
- Prolonged stasis during venopuncture
- □ Insufficient specimen
- □ Incorrect specimen container
- □ Inappropriate sampling site
- □ Incorrect specimen storage

# Analysing the specimen



#### Interpretation of results

- □ Value obtained with a particular parameter is interpreted as increased, decreased or within normal (reference) range
- Reference values: Values obtained from individuals who are in good health as judged by other clinical and laboratory parameters, after suitable standardization and statistical analysis, under definite laboratory conditions.
- Normal (Reference) Range: Values within which 95% normal healthy person's fall. The cut off values are set as mean reference value +/- N times standard deviation, of a normal healthy population; where N varies between 1, 2 and 3.

### **Quality Control**

- A major role of the clinical laboratory is the measurement of substances in body fluids or tissues. To fulfill these aims the data generated has to be **reliable** for which strict quality control has to be maintained.
- Quality control is defined as the study of those sources of variation, which are the responsibility of the laboratory, and the procedures used to recognize and minimize them.
- Quality control involves consideration of a reliable analytical method. Reliability of the selected method is determined by its accuracy, precision, specificity and sensitivity

## **Quality Control**

- Accuracy has to do with how close the mean of a sufficiently large number of determinations on a sample is to the actual amount of substance present and is dependent on the methodology used.
- □ **Precision** refers to the extent to which repeated determination on an individual specimen vary using a particular technique and is dependent on how rigorously the methodology is followed.
- □ **Specificity** is the ability of an analytical method to determine solely the analyte it is required to measure.
- **Sensitivity** is the ability of an analytical method to detect small quantities of the measured analyte.

### **Quality Control**

- Analytical methods **require calibration**, the process of relating the value indicated on the scale of the measuring device to the quantity required to be measured. Calibration is done using standard, the solution with which the sample is compared to arrive at the result.
- **Standard solutions** refer to the known amount of a substance in a solution in which its concentration is expressed in terms of moles or in weights per unit volume.

# Biological factors affecting the interpretation of results

- □ Sex of the patient.
- □ Age of the patient.
- □ Effect of diet.
- □ Time when sample was taken.
- □ Stress and anxiety.
- Posture of the patient.
- □ Effects of exercise.
- Medical history.
- □ Pregnancy.
- Drug history.

#### Case history 1

A blood specimen was taken from a 65-year-old woman to check her serum potassium concentration as she had been on thiazide diuretics for some time. The GP left the specimen in his car and dropped it off at the laboratory on the way to the surgery the next morning.

Immediately on analysing the sample, the biochemist was on the phone to the GP. Why?

Comment on page 152.