Antiepiletpic Drugs (AEDs) and sedative hypnotics

- Epilepsy is a chronic disorder characterized by recurrent seizures, which are finite episodes of brain dysfunction resulting from abnormal discharge of cerebral neurons
- Epilepsy is the most prevalent neurological disorder affecting more than 0.5% of the world's population

Epilepsy

- The etiology of epilepsy is largely unknown even though recent evidence suggests that it may have a genetic component associated with its disease development.
- Seizures: are symptoms of disturbed electrical activity in the brain characterized by episodes of abnormal, excessive, and synchronous discharge of a group of neurons within the brain that cause involuntary movement, sensation, or thought.
- > seizures may result from primary or acquired neurological disturbances of brain function as a result of an imbalance between excitatory and inhibitory processes in the brain.
- There are many possible causes of seizures including brain tumors or infections, head trauma, neurological diseases, systemic or metabolic disorders, alcohol abuse, drug overdose, or toxicities.

Classification of epileptic seizures

- Partial (local, focal) seizures
 - A. Simple (consciousness not impaired)
 - B. Complex partial seizures (psychomotor seizures)
 - Beginning as simple partial seizures, progressing to complex seizures
 - With impairment of consciousness at onset
 - C. Partial seizures evolving to secondarily generalized tonicclonic convulsions
- Generalized seizures (convulsive or nonconvulsive)
 - A. Absence seizures

Typical (petit mal)

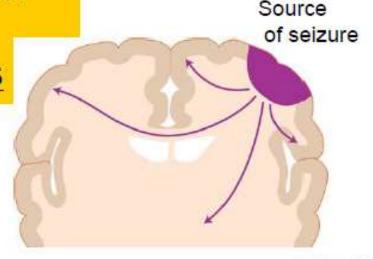
Atypical

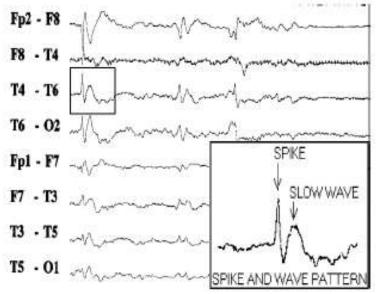
- B. Myoclonic
- C. Clonic
- D. Tonic
- E. Tonic-clonic (grand mal)
- F. Atonic
- Unclassified epileptic seizures (includes some neonatal seizures)

International Classification of Epileptic Seizures:

Partial Onset Seizures

- Simple Partial
- Complex Partial (consciousness is affected)
- Partial Seizures with secondary generalization





International Classification of Epileptic Seizures: Primary Generalized Seizures

- -Absence (Petit Mal)
- -Generalized Tonic+Clonic (Grand Mal)
- -Tonic
- -Atonic
- -Clonic and myoclonic

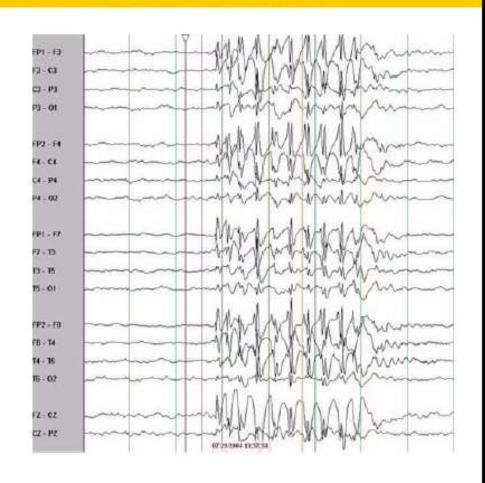
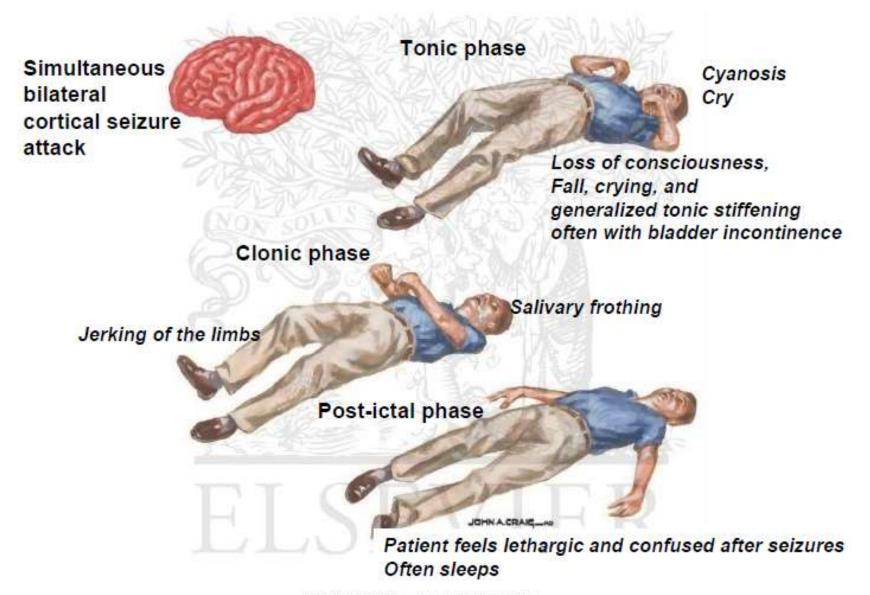




Figure 46-15 The patient was monitored with closed-circuit television simultaneous EEG and telemetry. The monitoring revealed stereotypical complex partial seizures. The patient is shown reading quietly in the period preceding the seizure

(A), during the period when she reported a feeling of fee (B), and during the period when there was alteration of a sciousness and an audible scream (C). (Courtesy of Dr M Salinsky, Oregon Health Sciences University Epilepsy Cer

Stereotypical complex partial seizures



Mechanisms of action of anticonvulsants

(a) modulation of voltage-gated ion channels (Na, Ca2, and K),

(b) Enhancement of aminobutyric acid (GABA)mediated inhibitory neurotransmission (c) attenuation of excitatory (particularly glutamate-mediated) neurotransmission in the brain.

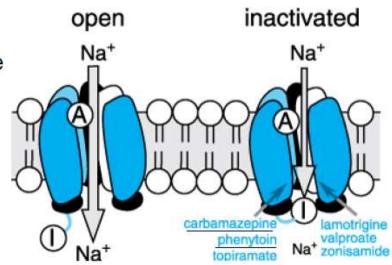
VOLTAGE-GATED SODIUM CHANNELS

in the presynaptic nerve terminal of the excitatory glutamate receptors

- 1-Phenyl-substituted succinimide cause some Na+-channel block
- 2-phenytoin, carbamazepine, oxacarbazepine, valproic acid, and felbamate is Phenobarbital, may also block voltage gated Na+-channel

Drugs which act on Na+ channel

- Phenytoin
- Carbamazepine
- Oxcarbazepine
- Lamotrigine



(2) Interaction with GABA_A receptor

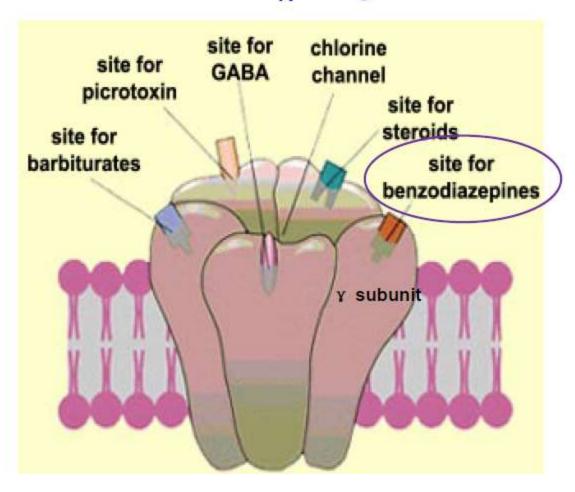
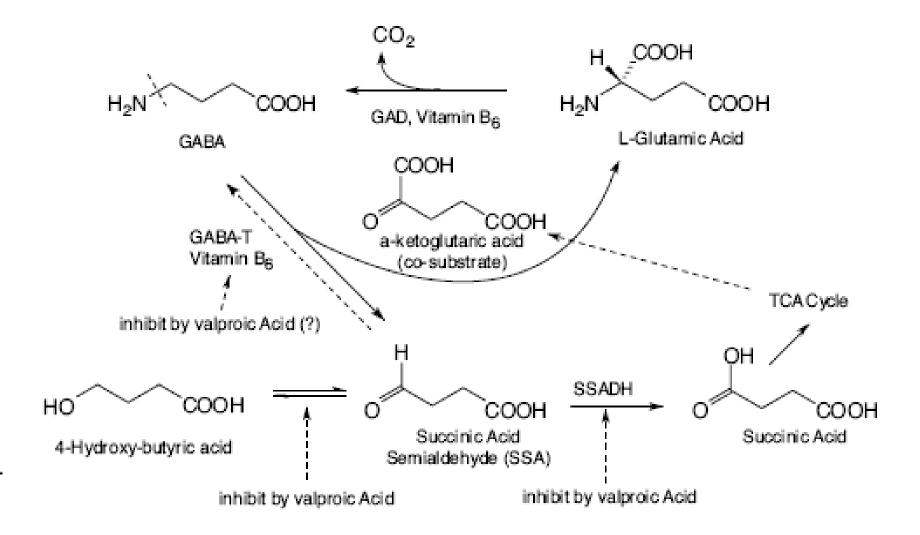
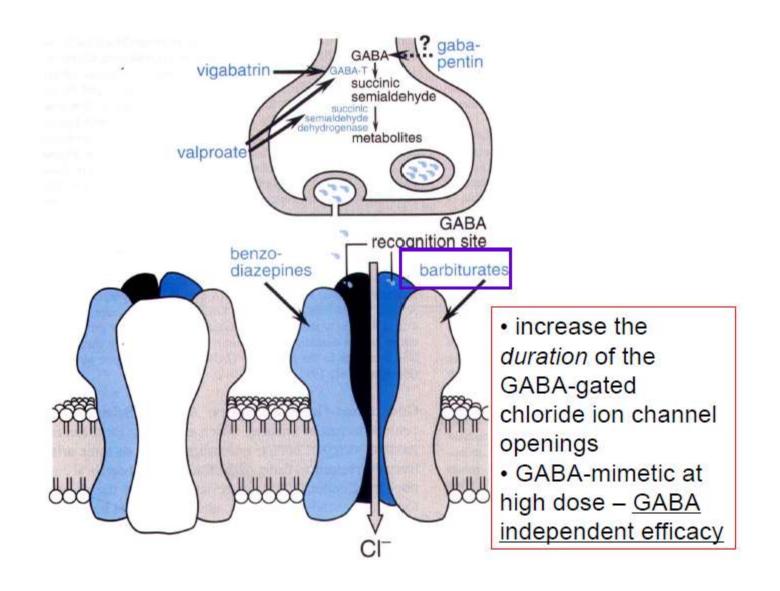


Figure 14.2 • Biosynthesis and metabolism of GABA.





GABAA Receptors as Targets for Anticonvulsants

The potential targets for AED's action on the GABAergic inhibitory synapses include:

- (a)drugs that enhance the biosynthesis of GABA (gabapentin, pregabalin, and VPA),
- (b) drugs that inhibit GABA degradation (vigabatrin),
- (c) drugs that inhibit the reuptake of GABA (tiagabine), and
- (d) drugs that bind to an allosteric site on the postsynaptic GABAA receptor complex that increase chloride conductance (barbiturates, BZDs, neurosteroids, felbamate FBM, Topiramate (TPM).

Antiepileptic General Structure

- Overall, R₁ and R₂ should be hydrocarbon
- Lower alkyls tend to be active against absence seizures and not active against generalized tonic-clonic or partial seizures
- If one of the hydrocarbon substituent is an aryl group activity tends to be directed toward generalized tonic-clonic and partial seizures and not absence seizures

Barbiturate

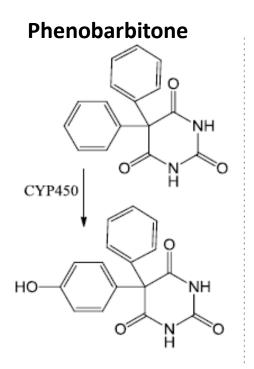
- The main antiepileptic drug is Phenobarbital major metabolite of which is the *p*-hydroxyl and/or the *p*-hydroxyglucuronide; about 25% excreted unchanged
- N¹ and N³ are not distinguishable.
- Both drugs being substituted with an aromatic ring at R₂ are effective against generalized tonic-clonic and partial seizures.

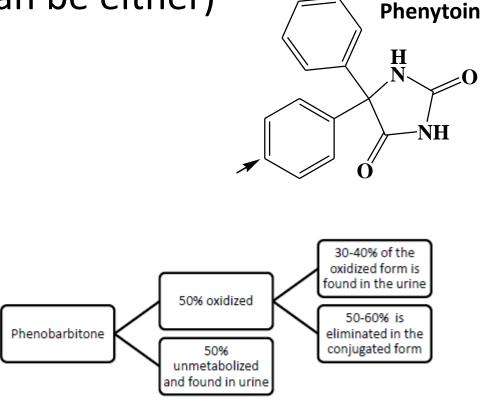
Phenobarbital

Conj. = glucuronide or sulfate

Example: Phenytoin

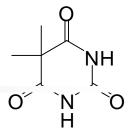
Here the oxidation will occur only at one aromatic ring (it can be either)







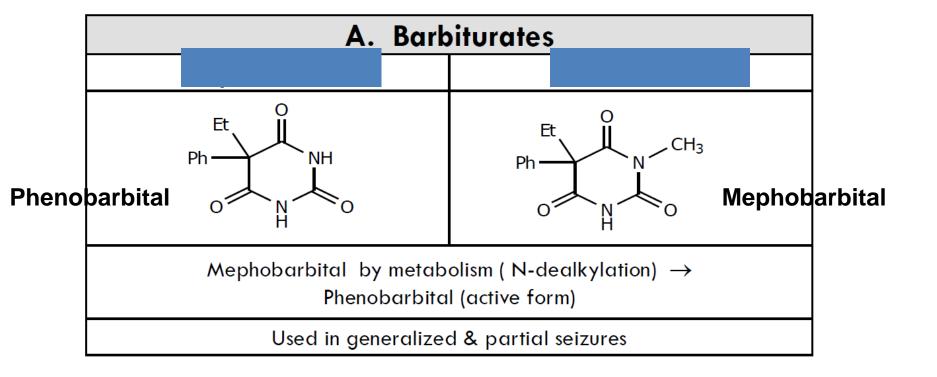
SAR of Barbiturates



- Both hydrogen atoms at C⁵ must be substituted
- There is a decrease in onset time and a decrease in duration as C⁵ alkyl chain length increases.
- Due to increasing lipid solubility increases rate of CNS penetration for shorter onset and increases susceptibility to microsomal metabolism due to penetration into hepatic cells
- **Common metabolic pathway is ω and ω-1 oxidation**
- Except for those with very high lipid solubility (thiobarbiturates), the barbiturates have short duration
- Bulk on C⁵ (i.e., aromatic ring) is a common feature for drugs with activity for generalized seizures and also for partial seizures and status epilepticus, but not good for absence seizures

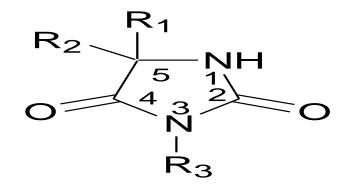
[i] Ureide derivatives

- <u>Classification</u>: A. Barbiturate. B. Hydantoins. C. Oxazolidinediones. D. Succinimides.
- General Structure:
- The general SAR:
- The substitution pattern at $\underline{C_5}$ of the hydantoins & oxazolidinediones or $\underline{C_2}$ of the succinimides determine the type of anti-convulsant activity.
- Hydantoins with <u>at least one C_5 phenyl gp</u> are the drug of choice in \rightarrow <u>generalized tonic-clonic seizures</u>
- The <u>diphenyl substitution</u> pattern <u>fpotency of anti-grand mal > single subs.</u>



Long acting (> 6 hr)	
Phenobarbital	Mephobarbital
Ph NH NH O H S-Ethyl-5-phenyl barbituric acid	Ph N CH ₃ N O H 5-Ethyl-1-methyl-5-phenyl barbituric
5 Emyr 6 phenyr barbhere dela	acid
Intermediate acting (3-6 hr)	Short acting (< 3 hr)
Aprobarbital	Secobarbital
NH NH O H 5-allyl-5-isopropyl barbituric acid	5-allyl-5-(1-methyl butyl) barbituric acid

Hydantoins



Hydantoins

- 1. Close structural relatives of barbiturates
- 2. 5-membered ring structure containing 2 nitrogen in ureide configuration
- 3. Only lacking the 6-oxo group and are cyclic monoacylureas rather than diacylureas

SAR of Hydantoins

Most of the clinically used drugs in this class possess bulky aromatic ring in position C⁵ that confers usefulness in generalized seizures, partial seizures and status epilepticus but not well for absence seizures

B. Hydantoins (imidazolidine-2,4-diones) **Phenytoin** Mephenytoin Ph Et Diphenyl - NH NH hydantoin 5,5-Diphenyl imidazilidine-2,4-dione 5-Ethyl-3-mehtyl-5-phenyl hydantoin

- Structurally close to barbiturates (differ in lacking of the 6oxo moiety)
- With <u>anti-generalized tonic clonic >>> Anti-absence.</u>

Hydantoin Drugs

Phenytoin is metabolized by *p*-hydroxylation followed by conjugation similar to Phenobarbital.

➤ **Mephenytoin** is the hydantoin analogue of mephobarbital which is also a prodrug, converted into the dealkylated derivative. Metabolism is also by *p*-hydroxylation and then glucuronidation

➤ Ethotoin is dealkylated to the active drug. In this case there is free hydrogen at C₅, which explains its very low potency. Metabolism is also by *p*-hydroxylation and then glucuronidation

Hypersensitivity reactions (idiosyncratic toxicity) to phenytoin and other aromatic AEDs in susceptible individuals are believed to stem from the reactions of these reactive intermediates(i.e., arene oxide, catechol, or *o*-quinone) with hepatic enzymes or other cellular proteins forming covalently bonded haptens.

- ➤ The reactive intermediate, arene oxide, is deactivated by either epoxide hydrolase to dihydrodiol (amajor urinary metabolite) or by the action of GSH and glutathione transferase.
- ➤ It has also been suspected that these reactive arene oxides or epoxides mediate the teratogenicity of phenytoin and other AEDs.

$\begin{array}{c|c} R_{1} \\ R_{2} \\ \hline 0 \\ \hline 0 \\ 4 \\ 3 \\ 0 \\ R_{3} \end{array}$

Oxazolidinedione

Oxazolidinediones

Replacement of the N-H group at position 1 of the hydantoin with an oxygen atom yields the oxazolidine-2,4-dione system

Trimethadione

Paramethadione

- ➤ **Trimethadione** is useful for absence seizures. Note the absence of bulky substituents at the C₅ position which are useful in absence seizures.
- ➤ It is metabolized to 5,5 dimethyl oxazolidine 2,4 dione (dimethadione) which is also active. Both trimethadione and dimethadione are excreted in the urine and are very toxic
- > Paramethadione is also N dealkylated, half life is 12-24 hours.

C. Oxazolidine2,4-Diones (Petit mal)

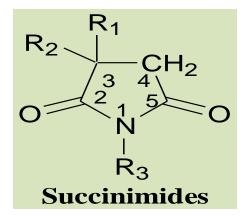
Trimethadione

,5,5-Trimethyl-1,3-oxazolidine-2-4-dione

- Prototype of the Anti-absence
- Produce <u>Aplastic anemia & bone marrow depression</u> → not used.

Succinimides

- This group of drugs resulted from a search for a less toxic version of the oxazolidinediones by replacing the "O" with CH₂
- work by blocking the low thresholdT-type calcium channels, thereby reducing the hyperexcitability of thalamic neurons that is specifically associated with absence seizure



D. Succinimides

Ethosuximide

2-Ethyl-2-methyl succinimide

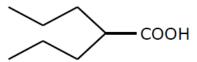
More effective & less toxic than trimethadione

 → the drug of choice for typical absence
 seizures.

[ii] Miscellaneous

Valproic acid (Depakin®)

Acetic a deriv



2-propyl petoic acid

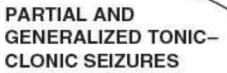
Metabolism:

- ① Direct conjugation (COOH)
- ② ω or ω -1 oxidation \rightarrow **5-hydroxy** & **4-hydroxy**

<u>derivative</u> [5-Hydroxy \rightarrow **2-n-propyl** glutamic acid.

Drug of choice for absence

(Maybe used against grand mal)



benzodiazepines, carbamazepine, ezogabine, lacosamide, oxcarbazepine, phenobarbital, phenytoin, pregabalin, fe primidone, rufinamide, tiagabine, topiramate, vigabatrin

felbamate

lamotrigine levetiracetam valproate

gabapentin

clonazepam

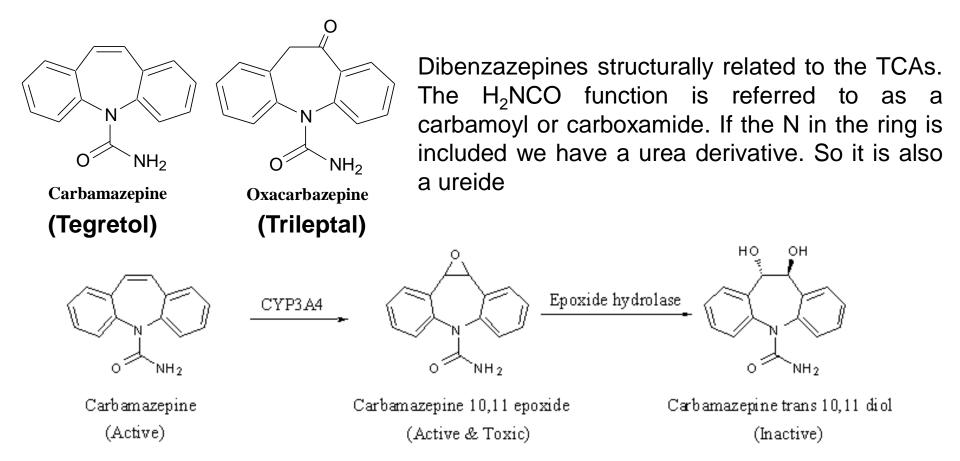
ABSENCE SEIZURES

ethosuximide, methosuximide, trimethadione

zonisamide

MYOCLONIC SEIZURES topiramate

Miscellaneous - Iminostilbenes



Oxcarbazepine does not undergo such epoxidation so is expected to be less toxic

Carbamazepine (Tegretol®)

Oxacarbazepine (Trileptal)

Drug of choice for grand mal

& used for partial not against petit

S.E: Bone marrow depression & aplastic anemia

Less side effect

No epoxide formation

Synthesis of Carbamazepine

Gabapantin acetic a deriv H₂N COOH GABA analogue but NOT acting on GABA receptors MOA: may involve L-amino acids transporter proteins.

Novel Broad-Spectrum Anticonvulsants

LAMOTRIGINE (LAMICTAL):phenyltriazine class found effective against refractory partial seizures.

mechanism of action: blockade of sodium channels that is both voltage- and use dependent.

It also inhibits the high-threshold calcium channel, possibly through inhibition of presynaptic N-type calcium channels and also blocks glutamate release

Lamotrigine

Anticonvulsants Acts on a Selective Molecular Target

TIAGABINE (GABITRIL)

- A glance at tiagabine's structure suggests an uptake inhibitor.
- Reportedly, it blocks GABA reuptake as a major mode of its anticonvulsant activity
- ➤ Nipecotic acid is a potent inhibitor of GABA reuptake into synaptosomal membranes, neurons, and glial cells. However, nipecotic acid fails to cross the blood-brain barrier following systemic administration because of its high degree of ionization.

Molecular Entity in 2009: Vigabatrin

- \blacksquare (±)-4-amino-5-hexenoic acid
- A GABA analog and is dosed as a racemic compound, with the S-enantiomer being the pharmacologically active form.
- The alkene group forms an irreversible, covalent bond with the gammaaminobutyric acid transaminase (GABA-T) and irreversibly inhibits it.
- The enzyme (GABA-T) is responsible for the metabolism of the inhibitory neurotransmitter GABA; its blockade leads to increased levels of GABA in the central nervous system.
- Thus, it is an antiepileptic drug indicated as a monotherapy for pediatric patients 1 month to 2 years of age with infantile spasms (IS) and as an adjunctive therapy for adult patients with refractory complex partial seizures (CPS) who have inadequately responded to several alternative treatments.
- It is essentially completely orally absorbed and widely distributed throughout the body. It is not significantly metabolized (80% of a dose is recovered as parent drug), although it does induce CYP2C9, and it is eliminated primarily through renal .

CNS DEPRESSANTS

CNS Depression

Sedation

Hypnosis

General Anesthesia

Poisoning

Death

SEDATIVE

Drugs that have an inhibitory effect on the CNS to the degree that they reduce:

- Nervousness
- Excitability
- Irritability without causing sleep
 (McKenry et al., 2003)

An effective **sedative** (anxiolytic) agent should reduce anxiety and exert a <u>calming effect</u> with little or no effect on motor or mental functions.

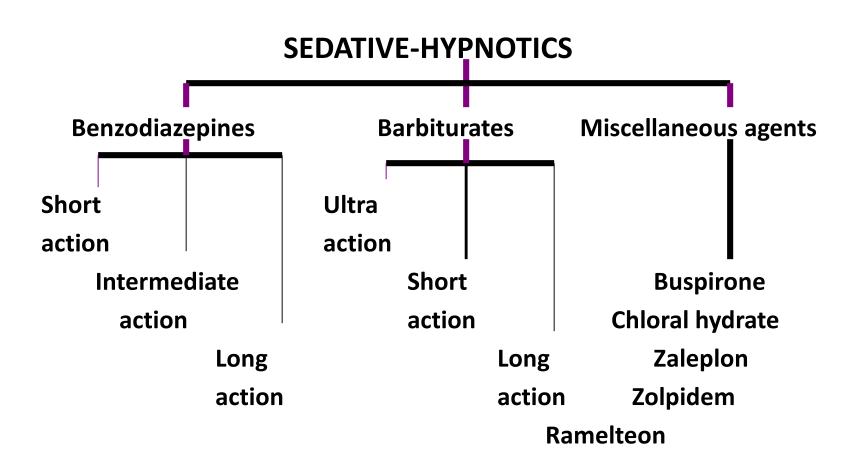
(Katzung et al., ed 11)

HYPNOTICS

- Calm or soothe the CNS to the point that they cause sleep
- A hypnotic drug should produce drowsiness and encourage the onset and maintenance of a state of sleep that as far as possible resembles the natural sleep state.
- A sedative can become a hypnotic if it is given in large enough doses → dose dependent

(Katzung; Goodman & Gilman)

SEDATIVE-HYPNOTIC DRUGS



[I] Sedative Hypnotics

- Action <u>from slight sedation to sleep</u> according to the <u>drug used</u>, <u>dose & its route of</u> <u>administration</u>
- <u>Uses of Sedatives</u>: emotional stress, hypertension, to control convulsion and adjunct to anesthesia.
- Uses of Hypnotics: To treat insomnia.

[i] Barbiturates

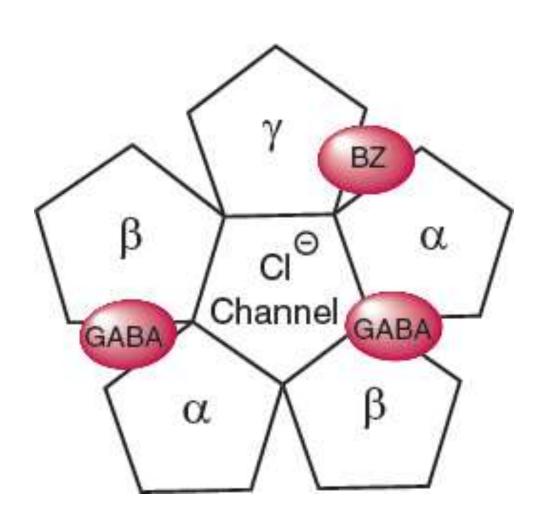
- Barbituric acid pKa = 4.12 (due to tautomerism→ ionized at physiological pH) → hydrophilic→ doesn't penetrate B.B.B.
- Monosubstituted barbituric acid at C5 has pKa =4.75
- Both barb.a'& monosubstituted barb.a' at physiological pH → found in ionic form→ doesn't penetrate B.B.B.→ inactive.
- **Disubstitution at C₅ has pKa= 7.6 + N-methylation pKa=8.4
- So, for good hypnotic activity of barbiturates we need:
- 1. Weak acid 2. ↑ Partition coefficient (to certain limit)
- <u>Prolonged use</u> → habituation & tolerance.
- Phenobarbital → Hepatic Microsomal Enzyme Inducing Drug [HME Inducer] → Tolerance + many Drug-Drug Interaction.

[II] Anxiolytics[i] Benzodiazepines (BZPs)

- Uses (drug of choice):
- Anxiolytics.
- Pre-anesthetic (induction of anesthesia)
- Hypnotic.
- Anticonvulsant.
- Central acting skeletal muscle relaxant.
- Advantage over barbiturates:
- <u>More safe</u> (overdose → <u>no respiratory depression</u>)
- Low tendency of drug interaction.

Disadvantage:

 Slow eliminated from body → due to formation of <u>active metabolite in</u> brain & liver → <u>hangover effect & accumulation on repeated dose.</u>



```
** BZPs potency depend on lipid solubility \rightarrow \uparrow lipophilicity will \uparrow potency \rightarrow (more distributed to the brain).

***BZPs duration \rightarrow depend on metabolism.
```

- MOA:
- Bind to <u>BZPs recognition site on BZPs receptor</u> \rightarrow <u>enhancing effect of GABA_A receptors</u> \rightarrow \uparrow Chloride ion flux into the neurons.

Serendipitous Discovery of Librium without a Lea

In 1955 Roche set out to prepare a series of benzheptoxdiazines as potential new tranquilizer drugs, but the actual structure was found to be that of a quinazoline 3-oxide.

No active compounds were found, so the project was abandoned

In 1957, during a lab cleanup, a vial containing what was thought to be the latter compound (X = 7-Cl, $R^1 = CH_2NHCH_3$, $R^2 = C_6H_5$) was sent for testing, and it was highly active.

Further analysis showed that the actual structure of the compound was the benzodiazepine 4-oxide, Librium, presumably produced in an unexpected reaction of the corresponding chloromethyl quinazoline 3-oxide with methylamine.

Librium

① 1,4-Benzodiazepine-4-oxides Chlordiazepoxide

Chlordiazepoxide is <u>the prototype & the most potent member</u>

It's Long acting drug due to formation of several active metabolites

② 1,4-Benzodiazepine-2-ones

Ring System

5-Phenyl-1,4-benzodiazepin-2-one

Ring A:

• <u>Electron withdrawing gp at C₇</u> is <u>essential</u> (Cl, Br, F, NO₂, CN) [the more electron attracting effect \rightarrow the more activity].

• $C_7 \rightarrow NO_2$ (intermediate duration) \rightarrow due to metabolism of NO_2 into NH_2 \rightarrow Acetylation (inactive metabolites).

Position 6,8,9 should not subistituted.

Ring B:

- The presence of **7-membered imino-lactam ring** is **essential**.
- The <u>2-carbonyl function</u> is <u>essential</u> for activity.
- The N_1 -substitution should be <u>small</u> \rightarrow <u>Xpt if active matabolits produced</u> as (flurazepam & prazepam)
- Alkyl subs at $C_3 \to \downarrow$ activity.
- The presence or absence of 3-OH is important pharmakokinetically.
 - * Without 3-OH \rightarrow non-polar (long duration).
- * With 3-OH \rightarrow more polar \rightarrow more easily excreted as glucuronides \rightarrow short duration.

- **COOH** at $C_3 \rightarrow prodrug$ with **long half life**.
- <u>C₅ phenyl gp</u> promotes activity.
- <u>Saturation</u> of double bond between C_4, C_5 <u>OR</u> its shift to $C_3, C_4 \rightarrow \downarrow$ activity.
- <u>Ring C</u>:
- The presence of <u>ortho or diortho substitution</u> with electron-attracting effect $\rightarrow \frac{7 \text{ activity.}}{}$
- Para substitution $\rightarrow \checkmark$ activity.

Diazepam (Valium®)	Nitrazepam	Clonazepam (Rivotril)
CH ₃	HN N	Anti-convulsant drug

Diazepam:

- •<u>PROTOTYPE</u> of this class. <u>Very non-polar \rightarrow rapidly absorbed \rightarrow Very potent.</u>
- •With long duration \rightarrow active metabolites.

Chlorazepate dipotassium

<u>Prodrug</u> \rightarrow activated by decarboxylation in stomach to Nordiazepam (active) \rightarrow with long $t_{1/2}$ (water sol \rightarrow IV)

③ 1,2-Annealated 1,4-Benzodiazepaines

• Short duration \rightarrow due to <u>rapid oxidation of -CH</u>₃ \rightarrow -CH₂OH \rightarrow conjugation & excretion.

5-Phenyl-1,4-benzodiazepin-2-one pharmacophore

- Nonbenzodiazepine GABA_A Agonists
- The discovery that there are several GABA_△ receptor subtypes (α_1 , α_2 , α_3 , and α_5) and that the α_1 -subtype is related most closely to benzodiazepine-induced sedation and hypnosis has led to the search for subtypeselective chemicals that would yield appropriate therapeutic outcomes. Currently there are three structurally distinct α_1 -subtype selective nonbenzodiazepines: zolpidem, eszopiclone, and zaleplon.

Zolpidem, Eszopiclone, and Zaleplon

Zolpidem

Effect:

- binds selectively to the BZ₁ subtype of BZ receptors and facilitates GABA-mediated neuronal inhibition
- Useful for the short-term treatment of insomnia
- Primarily a sedative (rather than an anxiolytic)
- are antagonised by flumazenil
- risk of tolerance and dependence < BZ

Pharmacokinetics:

- Rapidly absorbed in the GI tract following oral administration (75% reaches plasma)
- Metabolized in the liver and excreted by the kidney's
- Dosage reduction in hepatic dysfuction, elderly

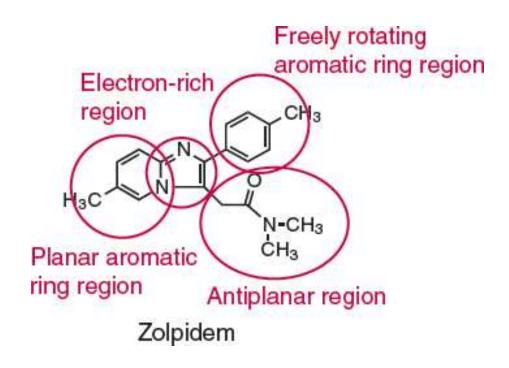
Structure—activity relationship studies revealed three activity regions necessary for binding to the GABA_A receptor (Fig. 8.9).

An Electron-Rich Planar Aromatic Ring Region

- Substitution with electronegative groups (e.g., chloro) decreases selective affinity for the α₁ subtype.
- Imidazole ring is necessary for selectivity since conversion of either of the imidazole nitrogens to hydrogen bond donors leads to loss of α_1 subtype selectivity.

Freely Rotating Aromatic Ring

• Substitution with electronegative groups (e.g., chloro) decreases selective affinity for the α₁ subtype.



Antiplanar Region

• The carbonyl group in this region can hydrogen bond with key residues in loop C of the α_1 subunit (Ser²⁰⁴ and Thr²⁰⁶/Gly²⁰⁷) and loop F of the γ_2 subuni

Zaleplon & Zopiclone

- Short half-life resembles zolpidem, $t_{1/2} = 1h$
- Rapid onset and short duration of action are favorable properties for those patients who have difficulty falling asleep.
- Only approx. 30% of an orally administered dose reaches the plasma, and most of that undergoes first-pass elimination
- Half as potent as zolpidem
- Improves sleep quality w/o rebound insomnia, and little chance of developing dependency

[ii] <u>Cyclopyrolone derivatives</u> Zopiclone

- Advantage:
- 1. *No withdrawal* symptoms.
- 2. No accumulation after repeated doses.
- 3. *Rapidly* induce sleep.

[ii] Miscellaneous

Buspirone (Buspar®)	Ondansetron (Zofran®)	
Aza spiro decane dione derivative	N N CH ₃	
 Uses: Anxiolytic & Antidepressant M.O.A: 5-HT_{1A} partial agonist. S.E.: Block Dopamine receptors → (EPS) 	- <u>Uses:</u> Anxiolytic, Anti-depressant & Anti-emetic & anti-psychotic <u>M.O.A:</u> <u>5-HT₃ antagonist</u> .	

Buspirone: 5 -HT1A -receptor agonists

• Pharmacokinetic:

- rapidly absorbed orally → extensive first-pass metabolism
- The elimination half-life of buspirone is 2–4 hours, and liver dysfunction may slow its clearance.
- Rifampin, an inducer of cytochrome P450, decreases the half-life of buspirone; inhibitors of CYP3A4 (eg, erythromycin, ketoconazole, grapefruit juice, nefazodone) can markedly increase its plasma levels.

Adverse effect:

- causes less psychomotor impairment than benzodiazepines
- does not potentiate effects of conventional sedative-hypnotic drugs
- elderly patients do not appear to be more sensitive to its actions.
- Nonspecific chest pain, tachycardia, palpitations, dizziness, nervousness, tinnitus, gastrointestinal distress, and paresthesias and a dose-dependent pupillary constriction
- FDA category B drug in terms of its use in pregnancy.

Differences between buspirone and benzodiazepines:

- 1- The full anxiolytic effect of buspirone takes several weeks to develop, whereas the anxiolytic effect of the benzodiazepines is maximal after a few days of therapy.
- 2- In therapeutic doses, buspirone has little or no sedative effect and lacks the muscle relaxant and anticonvulsant properties of the benzodiazepines.
- 3- Buspirone does not potentiate the central nervous system depression caused by sedative—hypnotic drug or by alcohol
- 4- Buspirone does not prevent the symptoms associated with benzodiazepine withdrawal.